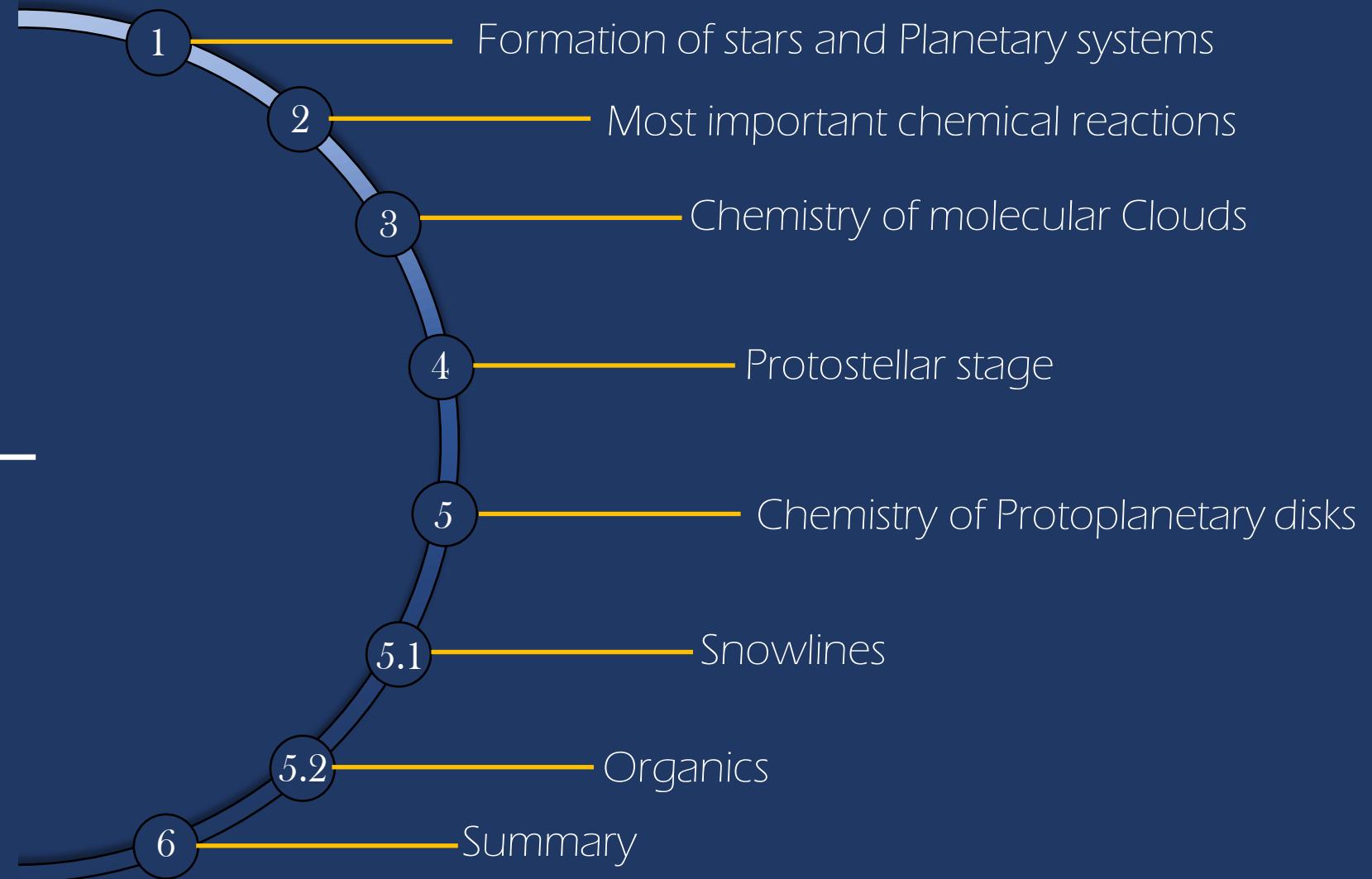


Astrochemistry and compositions of planetary systems

Daniel Ziegler 21.11.2023

Outline:



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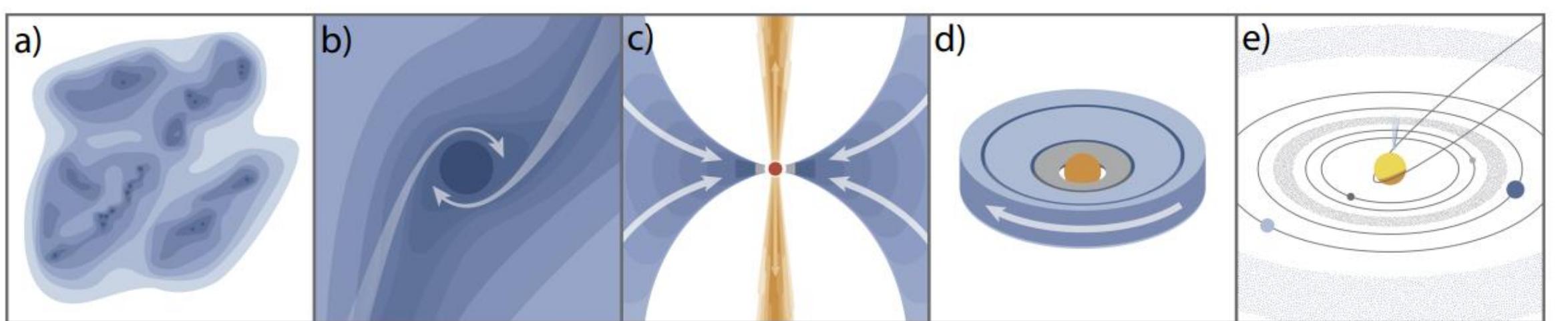
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Formation of stars and Planetary systems

Formation of stars and Planetary systems



Öberg and Bergin (2020) doi:10.48550/arXiv.2010.03529

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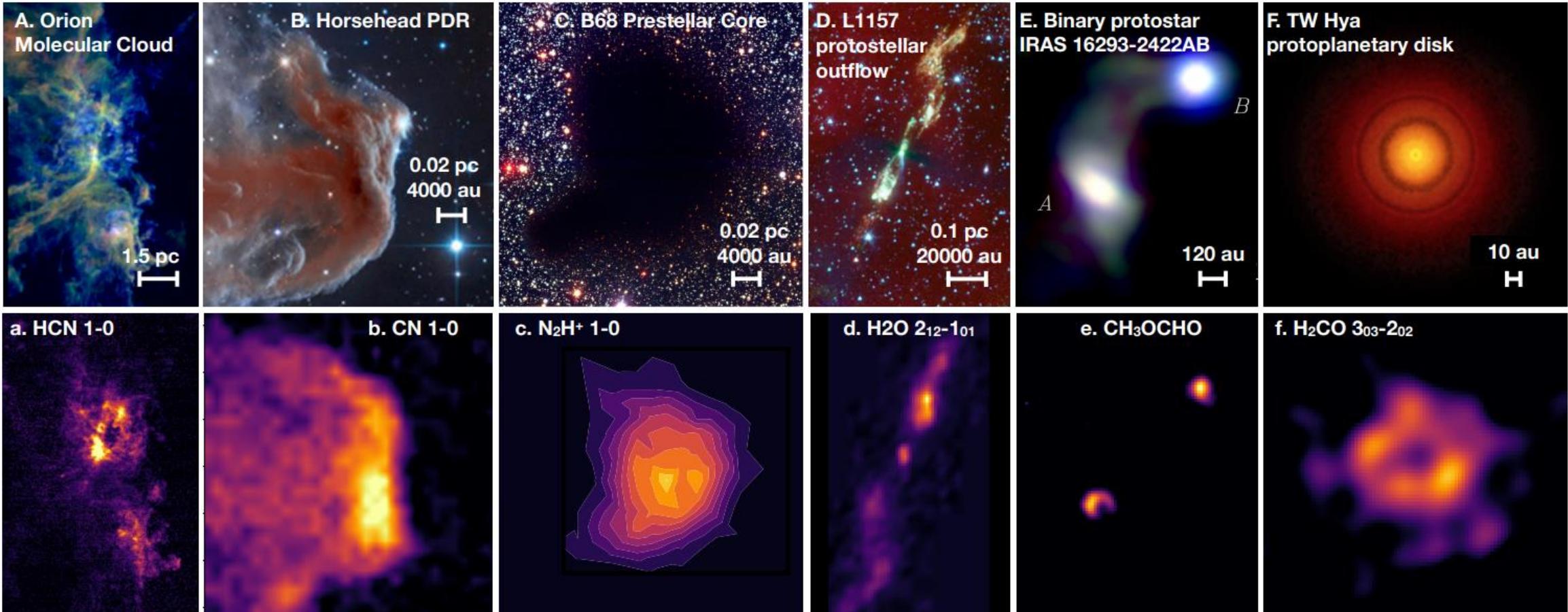
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Chemical structures of different stages of star and planet formation



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Most important chemical reactions

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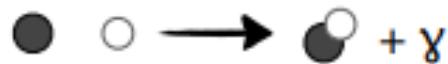
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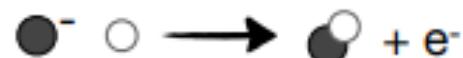
Gas phase chemistry

Bond formation

Radiative association



Associative detachment



Bond destruction

Photo-dissociation

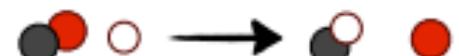


Dissociative recombination



Bond rearrangement

Ion-molecule, neutral-neutral or charge transfer



Öberg and Bergin (2020) doi:10.48550/arXiv.2010.03529

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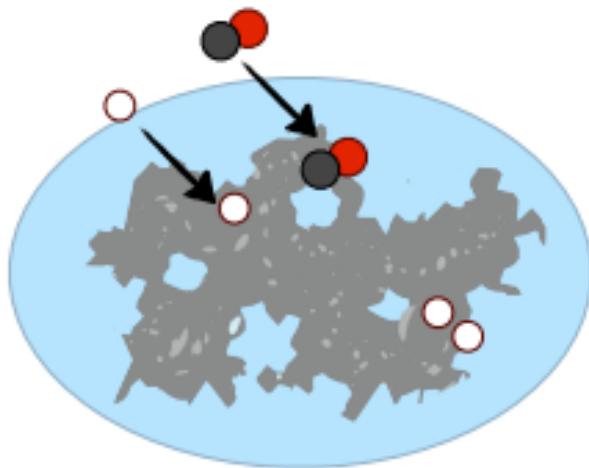
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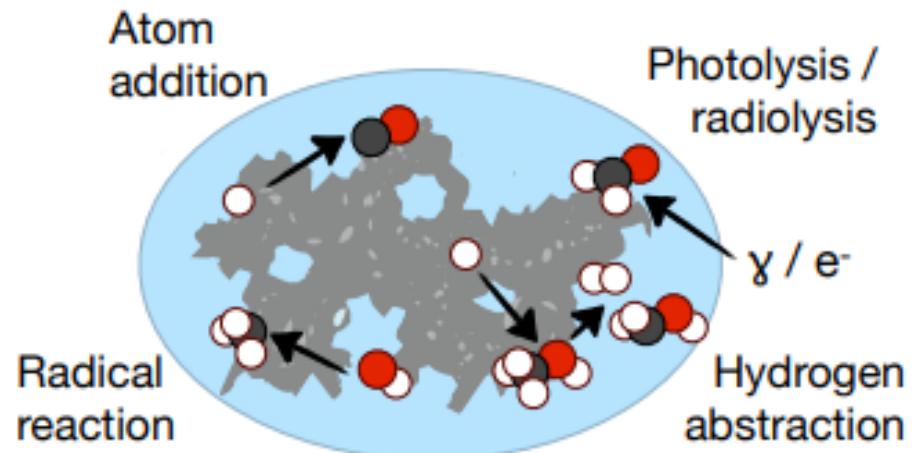
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Grain-surface and ice processes

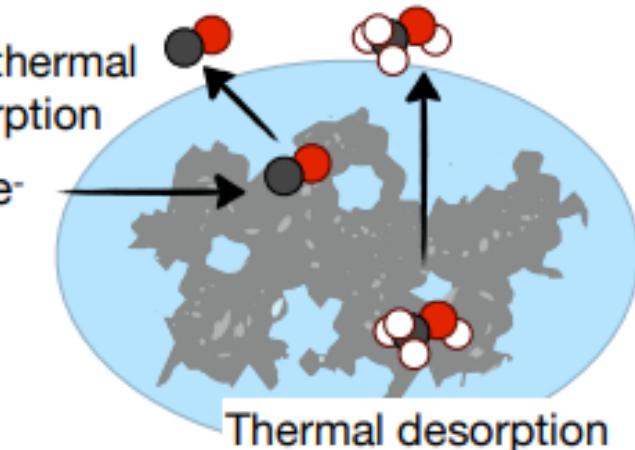
Freeze-out / adsorption



Grain surface and ice chemistry



Ice sublimation / desorption



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Classification of volatiles

- Hypervolatiles: CH_4 , CO and N_2
- Volatiles: H_2O , CH_3OH , CO_2 and NH_3
- Refractories: silicates and carbonaceous grains

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Chemistry of molecular Clouds

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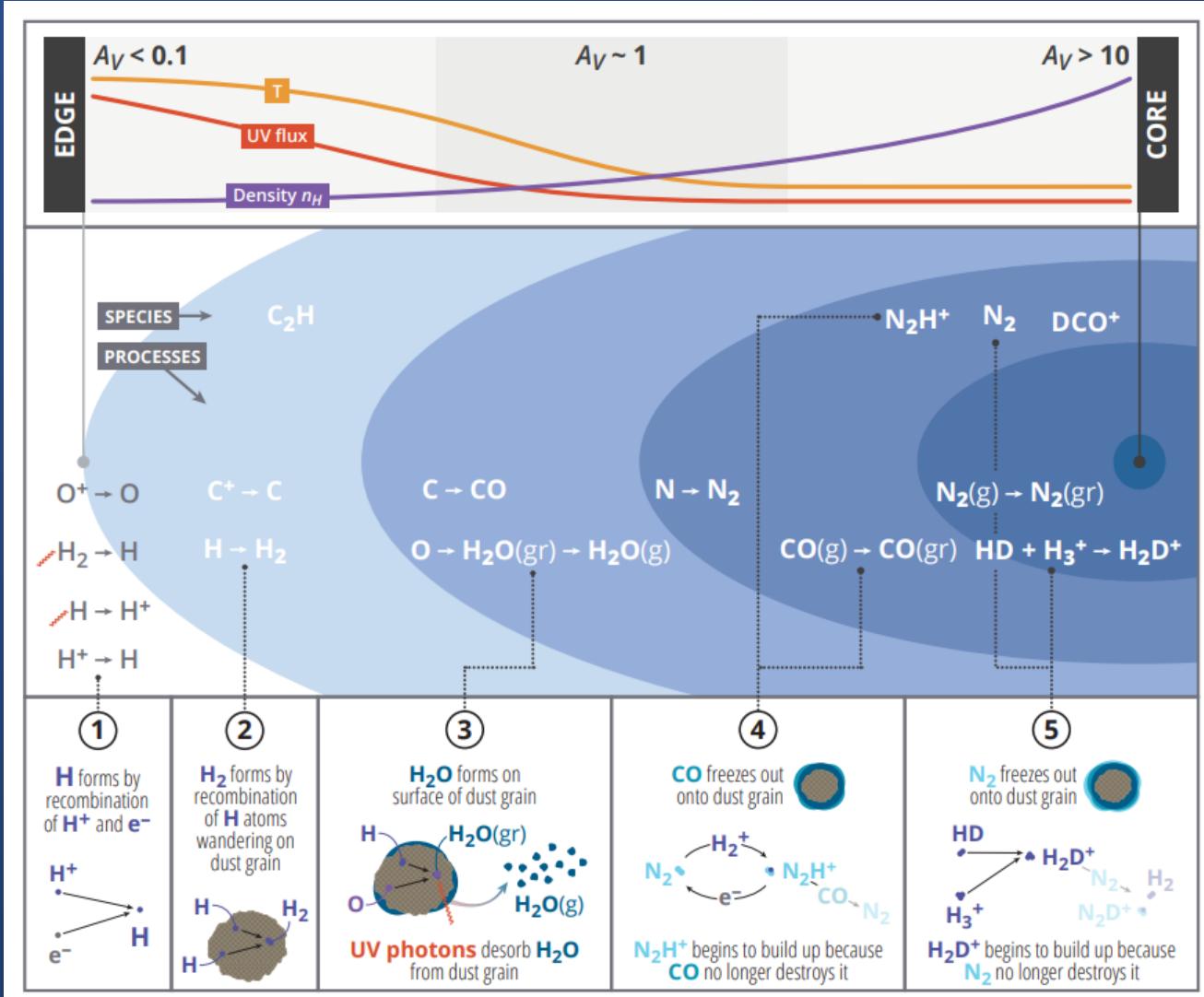
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Chemical structure of molecular clouds

- Radiation field
- Gas temperature
- Dust temperature
- Gas density

Chemical structure of molecular clouds



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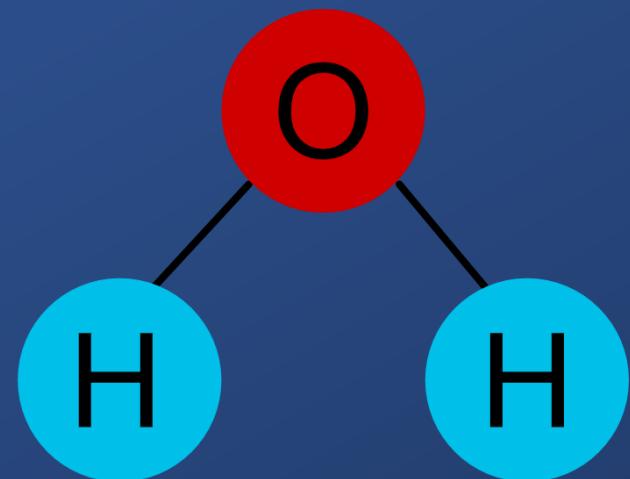
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The major molecules

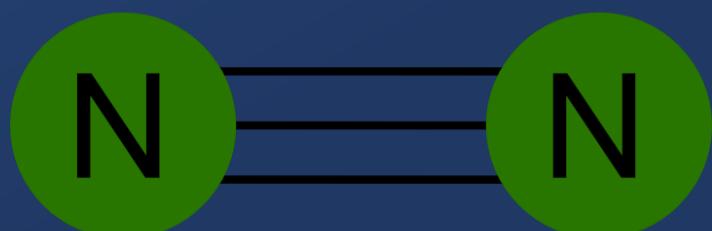
H_2O



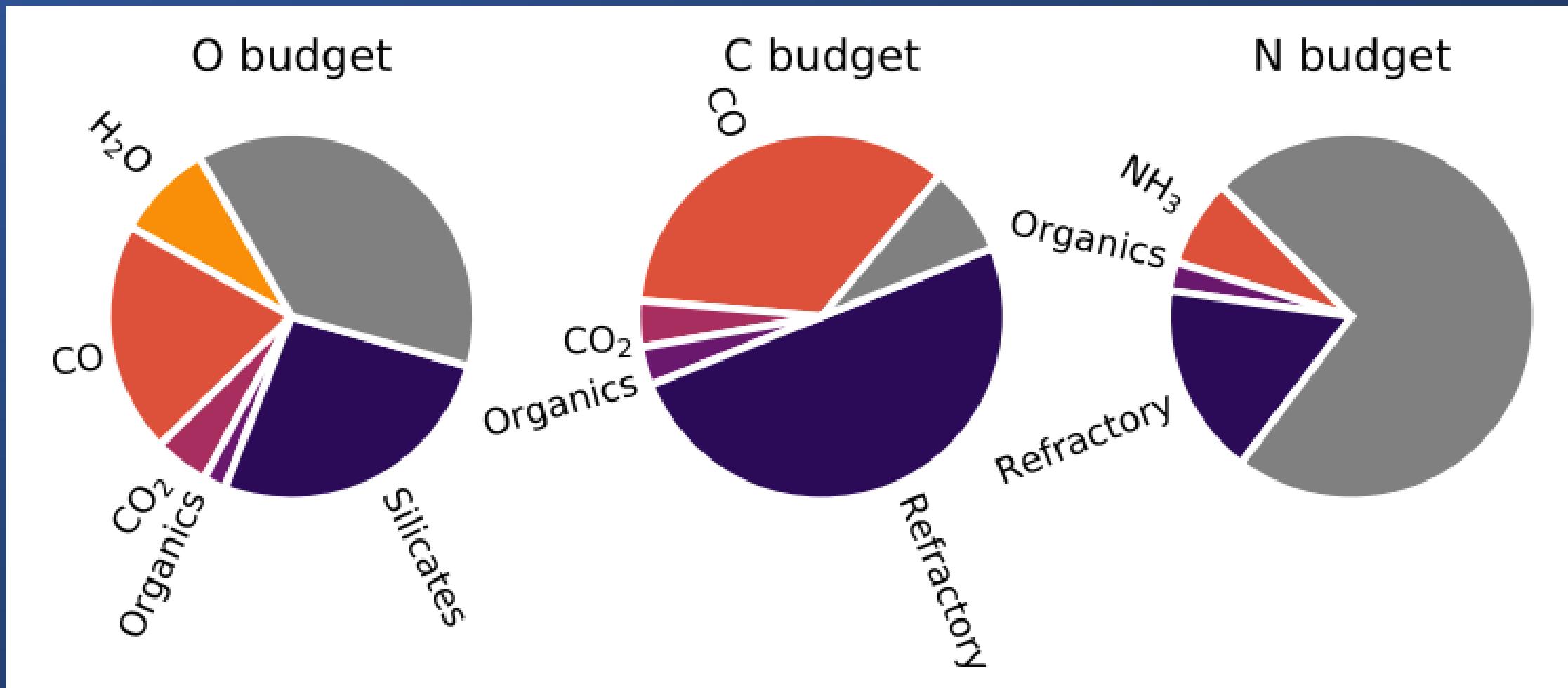
CO



N_2



O/C/N budgets



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Organic reservoir

- First generation of organic molecules
- Formed through gas-phase and ice-phase chemistry
- Most common organic molecules: CH_3OH and CH_4
- Some gas-phase reactions to form HCN and unsaturated organics
- Some complex organic molecules

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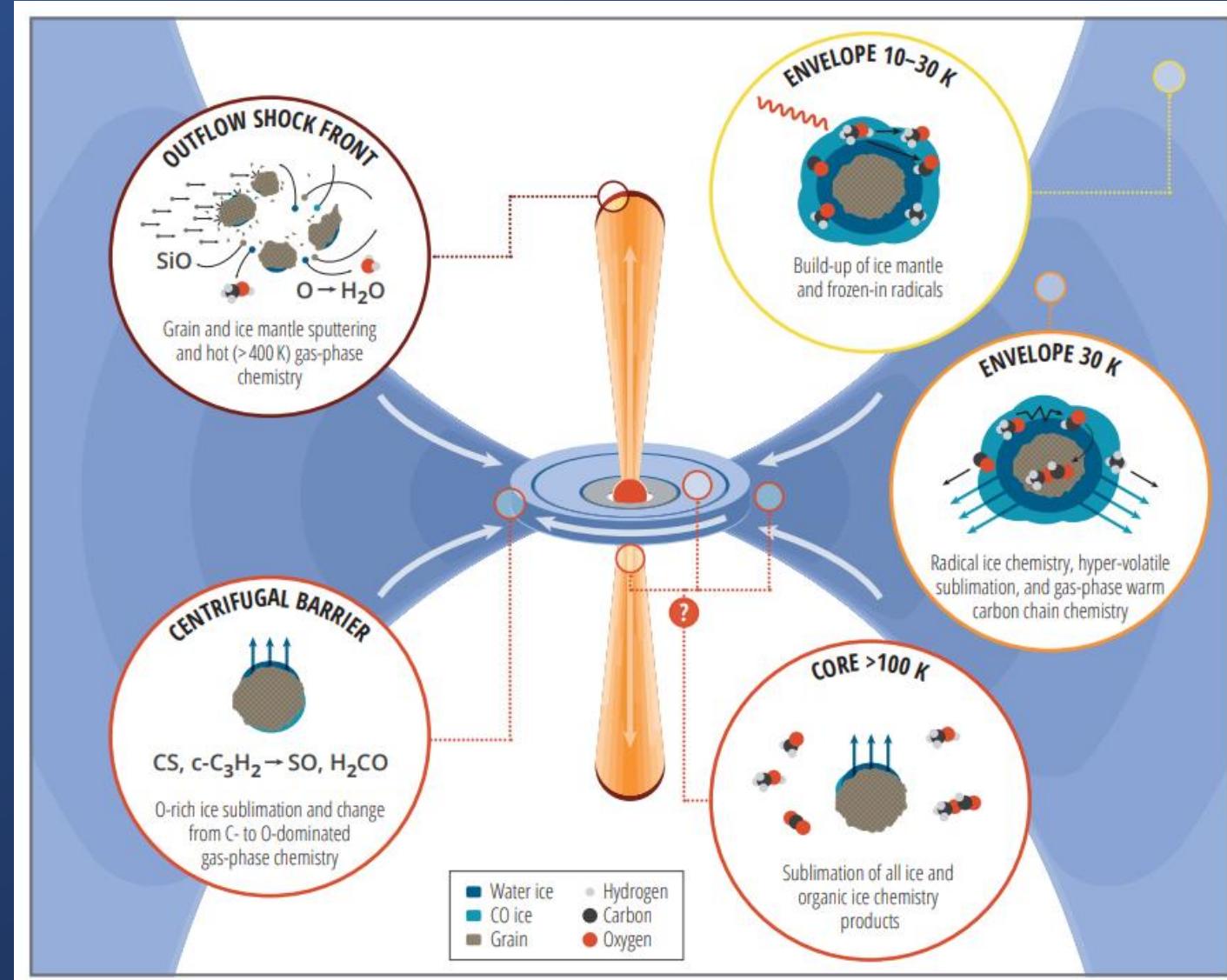
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Protostellar stage

Chemical structure of Solar-type protostars



Chemical structure of Solar-type protostars

- 25-30K : Most volatile ice constituents begin to sublime
- → Warm carbon chain chemistry
- Ice chemistry for more complex organic material
- 100-200K: We get hot corinos

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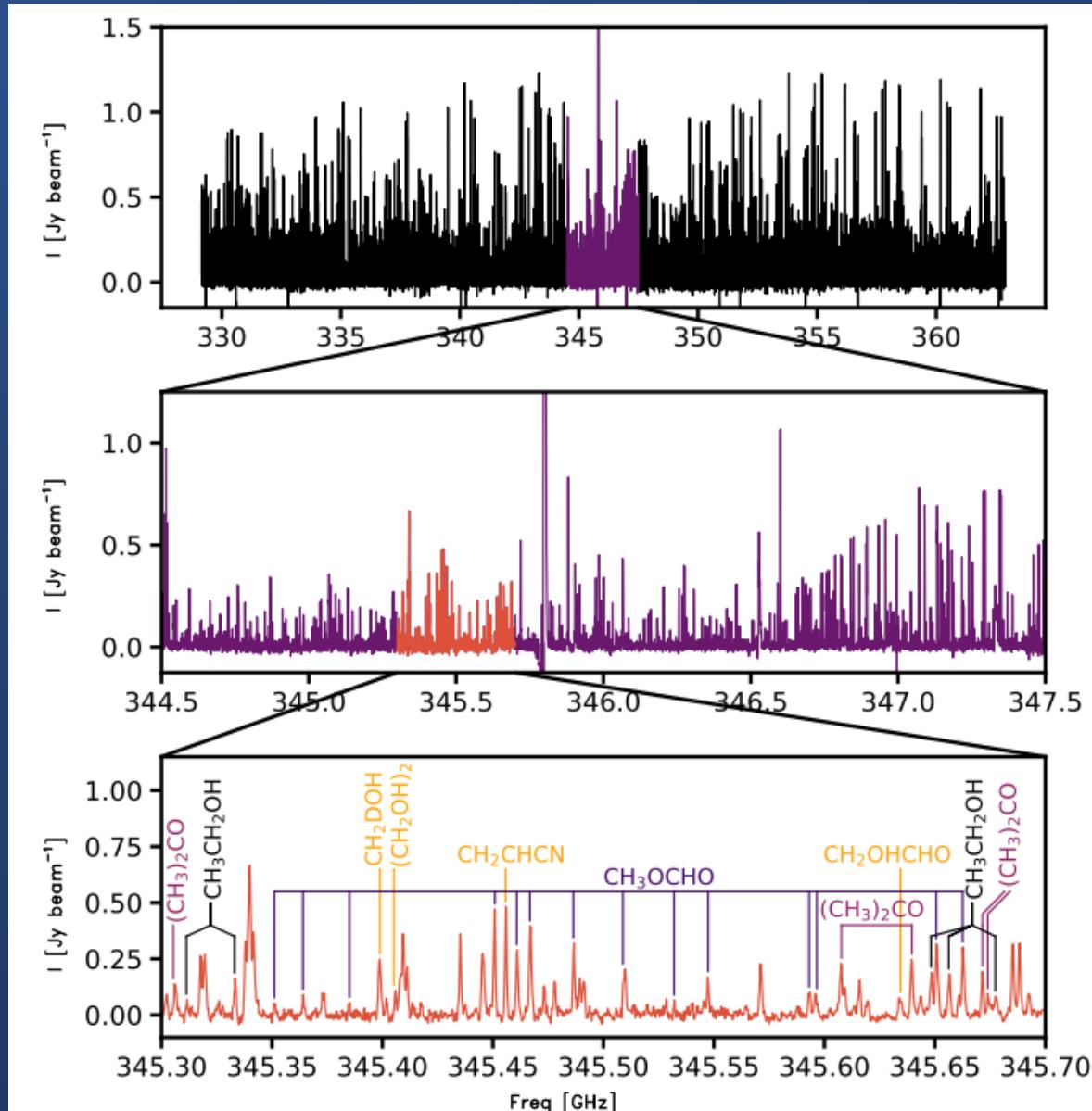
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Chemical structure of Solar-type protostars



Chemical structure of Solar-type protostars

- 25-30K : Most volatile ice constituents begin to sublime
- → Warm carbon chain chemistry
- Ice chemistry for more complex organic material
- 100-200K: We get hot corinos
- Chemistry mainly takes place in icy grain mantles
- Main elemental carriers at this stage remain the same

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Chemistry of Protoplanetary disks

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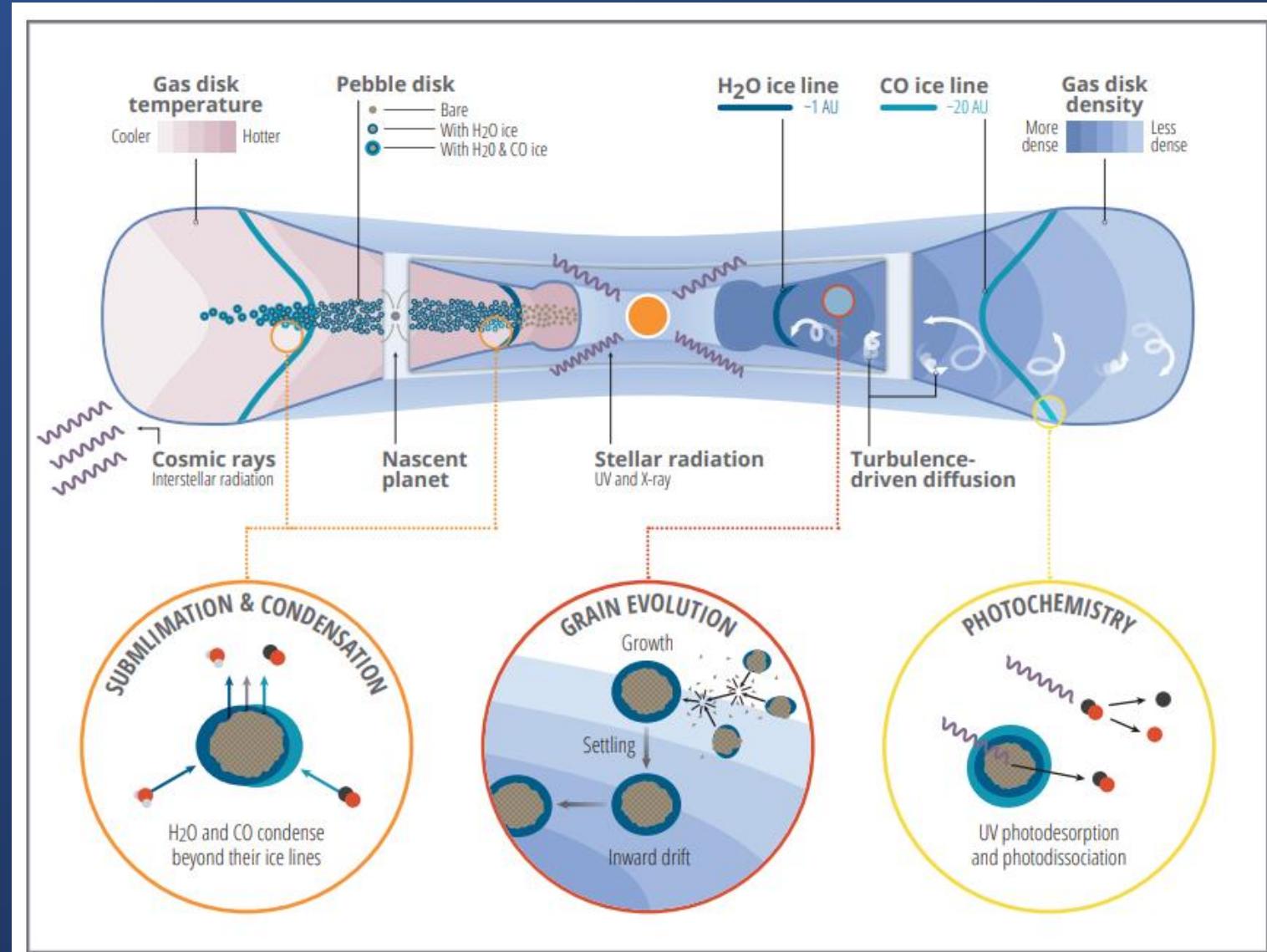
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Protoplanetary disk chemistry and planet formation

- Time dependent radial and vertical distribution of molecules
- Division of molecules between gas and solids
- Snowlines
- Grain compositions
- Grain fragmentation
- Coagulation properties
- → Pebble size distribution → Planetesimal and planet growth

Protoplanetary disk chemistry and planet formation



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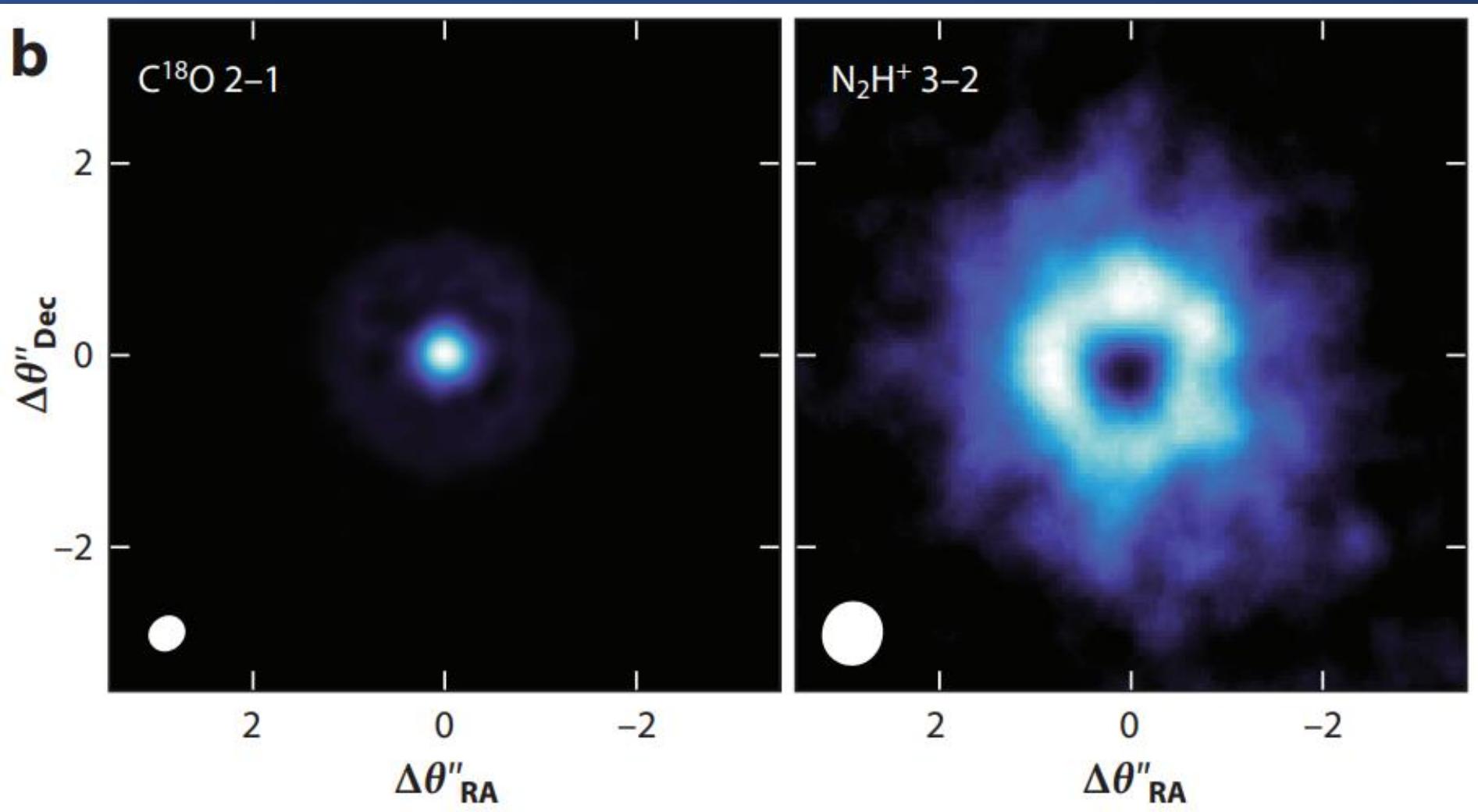
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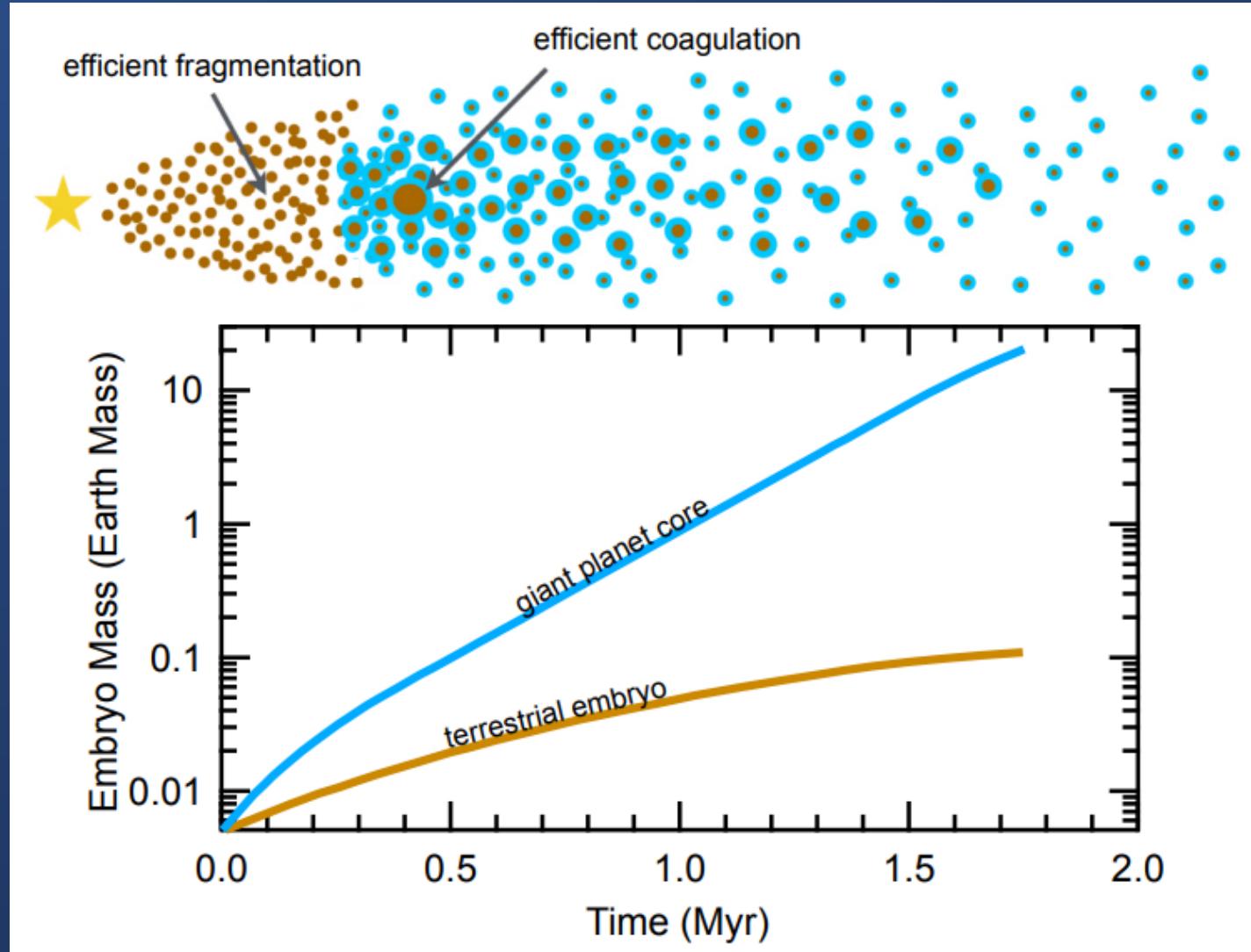
Snowlines

- Determine the elemental content of solids and gas at different disk locations
- Surface density increases
- Diffusive flows → Increasing column density of solids → Larger particles
- Properties changes across snowlines
- Localized regions of enhanced pressure

Snowlines

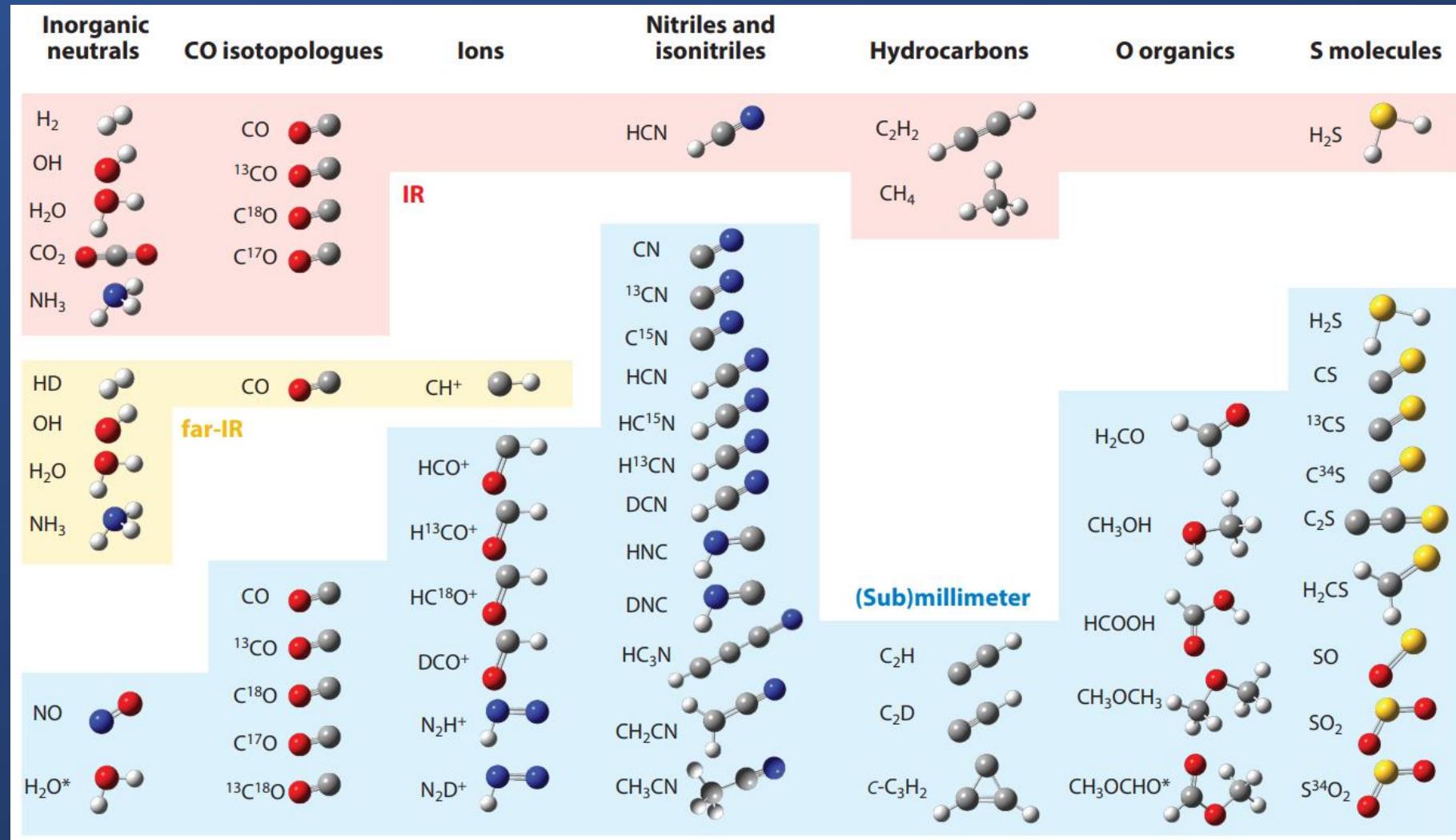


Growth of planets



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Disk organic chemistry



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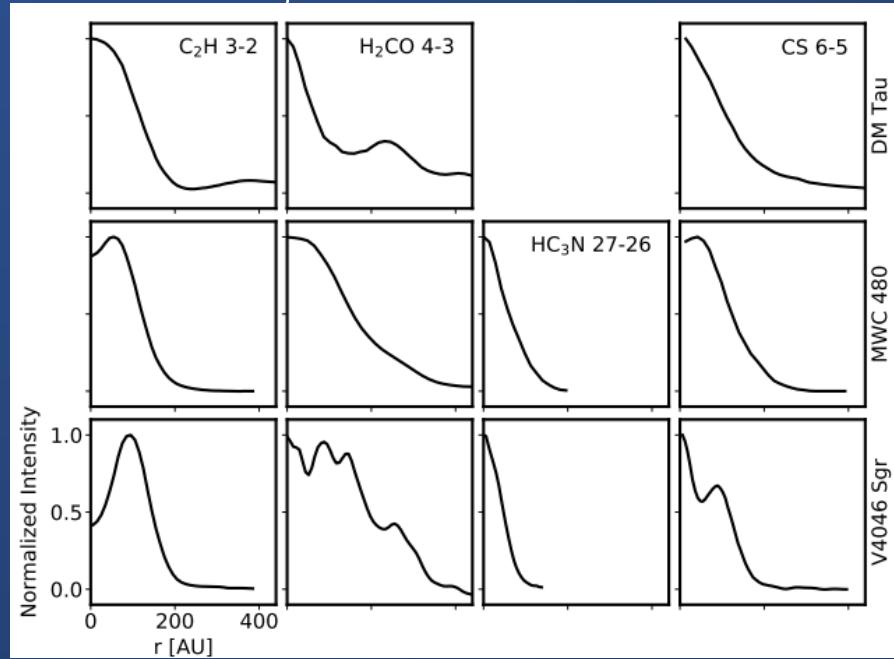
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Disk organic chemistry

- Small organic molecules in the inner disk atmosphere



Öberg and Bergin (2020) doi:10.48550/arXiv.2010.03529

- → Planetesimal organic compositions will strongly depend on both where and around which star the planetesimals formed.

Delivery of water and organics to temperate planets

- Three theories for the origin of water on a dry world:

1. In situ
2. Snowline evolution
3. Dynamical supply

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Summary

- Most abundant volatile elements: C, O, N
- In cloud cores: gas phase formation shifts to grain surface
- Most important molecules: CO, N_2, H_2O
- Organic feed-stock molecules are formed in cold cloud regions
- Snowlines play an important role in planet formation
- Unclear delivery of water and organics to temperate planets

Sources:

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