

# Sternentstehung - Star Formation

Winter term 2024/2025

Henrik Beuther, Thomas Henning & Caroline Gieser

15.10	<i>Today: Introduction &amp; Overview</i>	(Beuther)
22.10	<b>Physical processes I</b>	<b>(Beuther)</b>
29.10	--	
05.11	Physical processes II	(Beuther)
12.11	Molecular clouds as birth places of stars	(Beuther)
19.11	Molecular clouds (cont.), Jeans Analysis	(Henning)
26.11	Collapse models I	(Beuther)
03.12	Collapse models II	(Beuther)
10.12	Protostellar evolution	(Gieser)
17.12	Pre-main sequence evolution & outflows/jets	(Henning)
07.01	Accretion disks I	(Henning)
14.01	Accretion disks II	(Henning)
21.01	High-mass star formation, clusters and the IMF	(Gieser)
28.01	Extragalactic star formation	(Henning)
04.02	Planetarium@HdA, outlook, questions	
11.02	Examination week, no star formation lecture	(Beuther, Gieser, Henning)

**Book: Stahler & Palla: The Formation of Stars, Wileys**

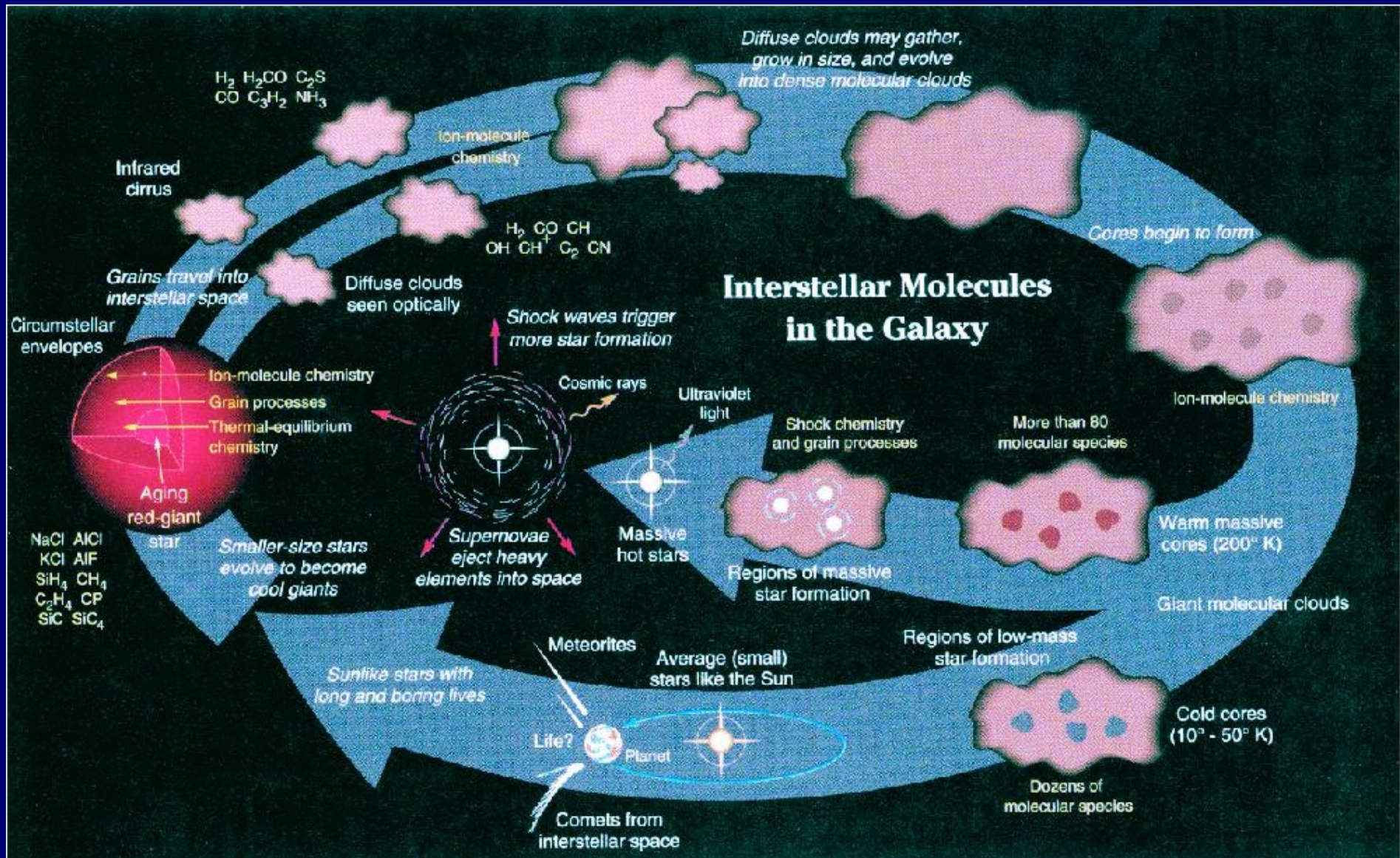
More Information and the current lecture files: [http://www.mpia.de/homes/beuther/lecture\\_ws2425.html](http://www.mpia.de/homes/beuther/lecture_ws2425.html)  
[beuther@mpia.de](mailto:beuther@mpia.de), [henning@mpia.de](mailto:henning@mpia.de), [gieser@mpia.de](mailto:gieser@mpia.de)

# Topics today

- The ISM, molecules and depletion
- Heating and cooling
- Radiation transfer and column density determination



# The cosmic cycle



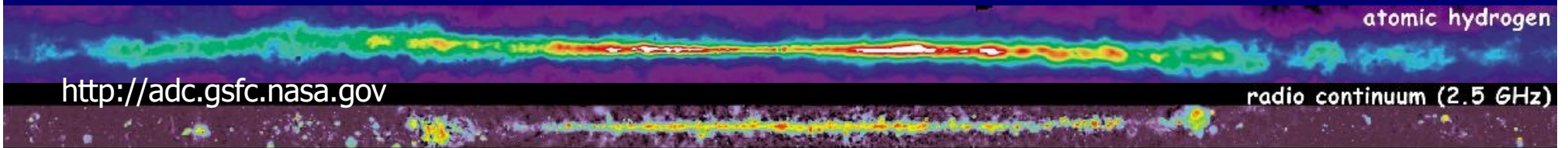
# Properties of Molecular Clouds

Type	n [cm <sup>-3</sup> ]	Size [pc]	T [K]	Mass [M <sub>sun</sub> ]
Giant Molecular Cloud	10 <sup>2</sup>	50	15	10 <sup>5</sup>
Dark Cloud Complex	5x10 <sup>2</sup>	10	10	10 <sup>4</sup>
Individual Dark Cloud	10 <sup>3</sup>	2	10	30
Dense low-mass cores	10 <sup>4</sup>	0.1	10	10
Dense high-mass cores	>10 <sup>5</sup>	0.1-1	10-30	100-10000

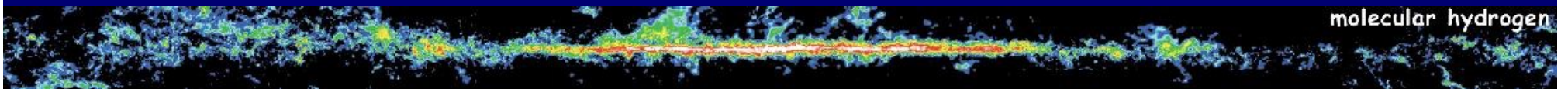


# Basics

Neutral and ionized medium



Stars form in the dense molecular gas and dust cores



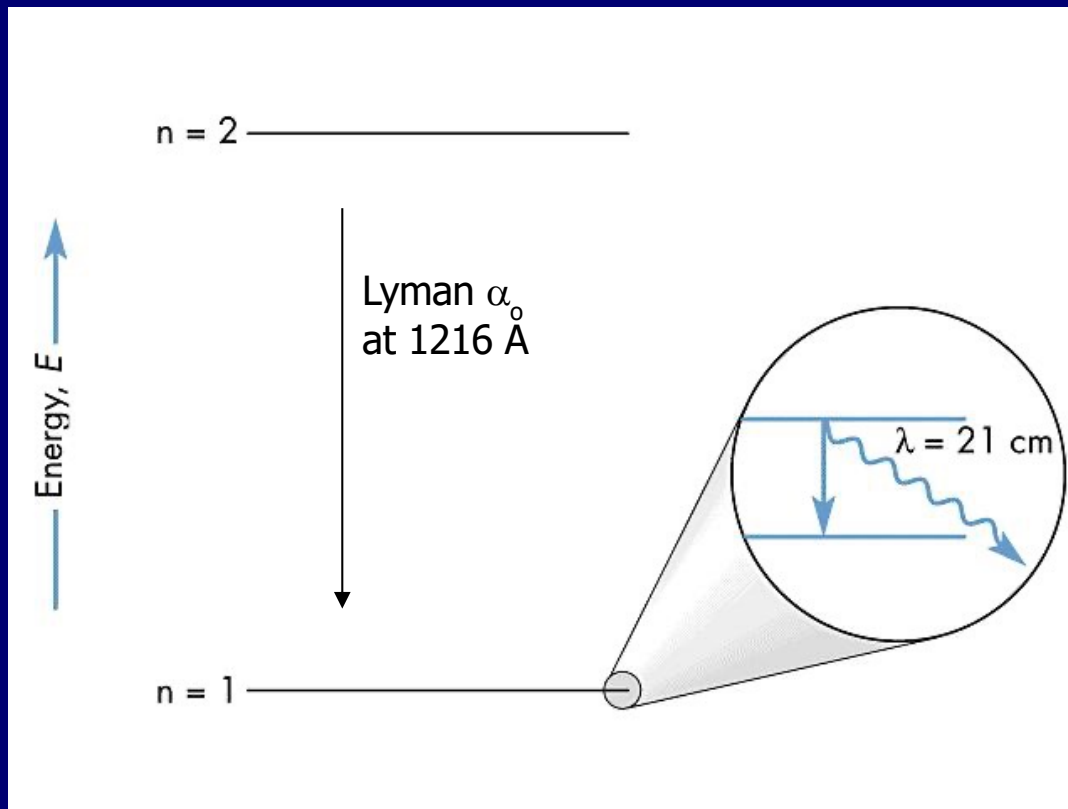
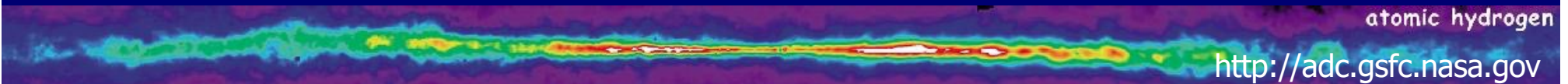
## **Most important astrophysical tools:**

Spectral lines emitted by various molecules

Absorption and thermal emission from dust

# The Interstellar Medium I

## Atomic Hydrogen



21cm line:  
electron spin  $S$  flip from parallel  
( $F=1$ ) to antiparallel ( $F=0$ )  
compared to the Proton spin  $I$ .

$$\Delta E = 5.9 \times 10^{-5} \text{ eV}$$

# The Interstellar Medium I

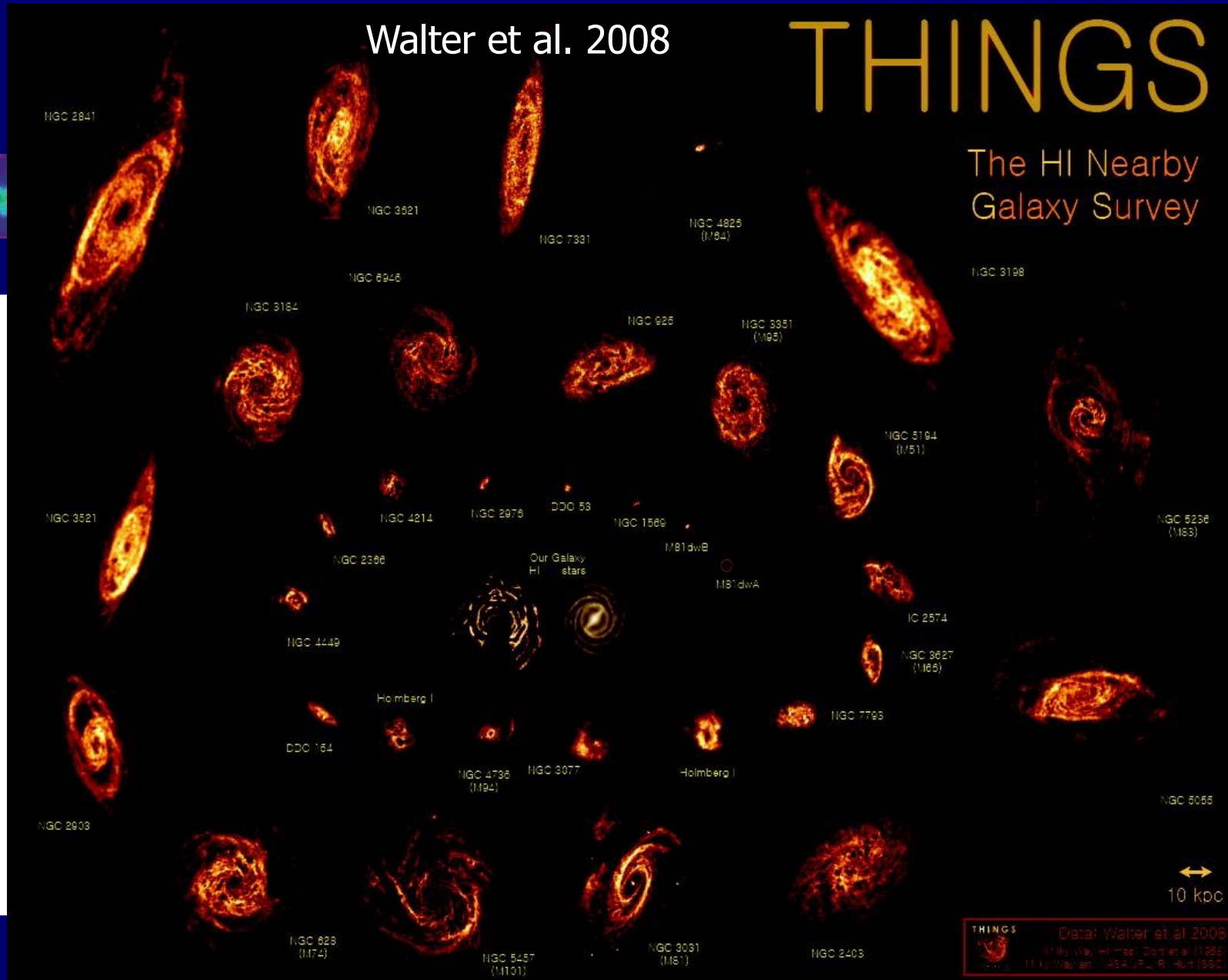
Walter et al. 2008

# THINGS

The HI Nearby  
Galaxy Survey

atomic hydrogen  
sfc.nasa.gov

Energy,  $E$  ↑



parallel  
))  
spin I.

# The Ionized gas

## Ionized gas

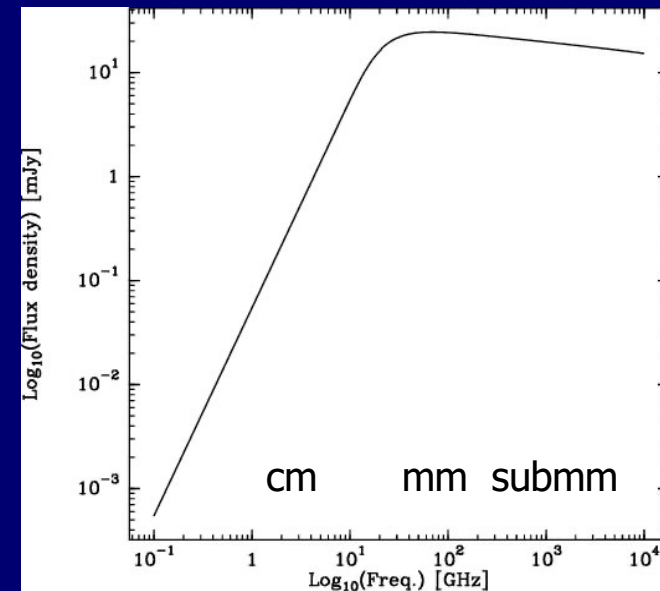
<http://adc.gsfc.nasa.gov>

radio continuum (2.5 GHz)

- Hydrogen recombination lines from optical to cm wavelengths
- Emission lines from heavier elements --> derive atomic abundances

He/H	0.1
C/H	$3.4 \times 10^{-4}$
N/H	$6.8 \times 10^{-5}$
O/H	$3.8 \times 10^{-4}$
Si/H	$3.0 \times 10^{-6}$

- Free-free emission between  $e^-$  and  $H^+$





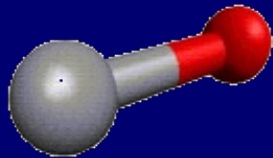
# The Molecular ISM

## Molecular Hydrogen

<http://adc.gsfc.nasa.gov>

molecular hydrogen

Carbon monoxide CO

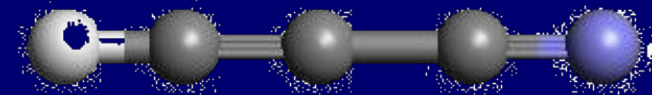


Formaldehyde H<sub>2</sub>CO



<http://www.cdms.de>

Cyanoacetylene HC<sub>3</sub>N



Excitation mechanisms:

- Rotation --> usually cm and (sub)mm wavelengths
- Vibration --> usually submm to FIR wavelengths
- Electronic transitions --> usually MIR to optical wavelengths

# Molecular ISM Basics

## History:

- Late 1930s: Detection of CH, CH<sup>+</sup> and CN in diffuse clouds by absorption of optical light from background stars
- 1960s: Detection of OH, NH<sub>3</sub> and H<sub>2</sub>O at radio wavelength
- 1970: CO

## Formation of molecules is an energy problem:

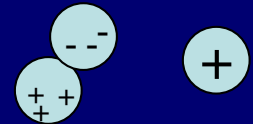
Two atoms approach each other with positive total energy  
→ rebound if no energy can be given away

## Possibilities:

- Simultaneous collision with 3rd atom carrying away energy  
--> unlikely at the given low densities
- Form a molecule in excited state, and then radiating away energy  
--> probability of such radiative association low as well

# Molecular ISM Basics

- Ion-molecule or ion-atom reactions can solve energy problem
- Neutral-neutral reactions on dust grain surfaces (catalytic) important
- Ion induces dipole moment in atom or molecule
  - > creates electrostatic attraction between the two.
  - > effective cross section increases over geometric values
- At low temperatures such reactions account for large fraction of molecules.
- However, not enough ions to account for large H<sub>2</sub> abundances
  - > grain surface chemistry important
- Simple molecules like CO or CS → ion-molecule chemistry,
- More complex molecules → grain surface chemistry important



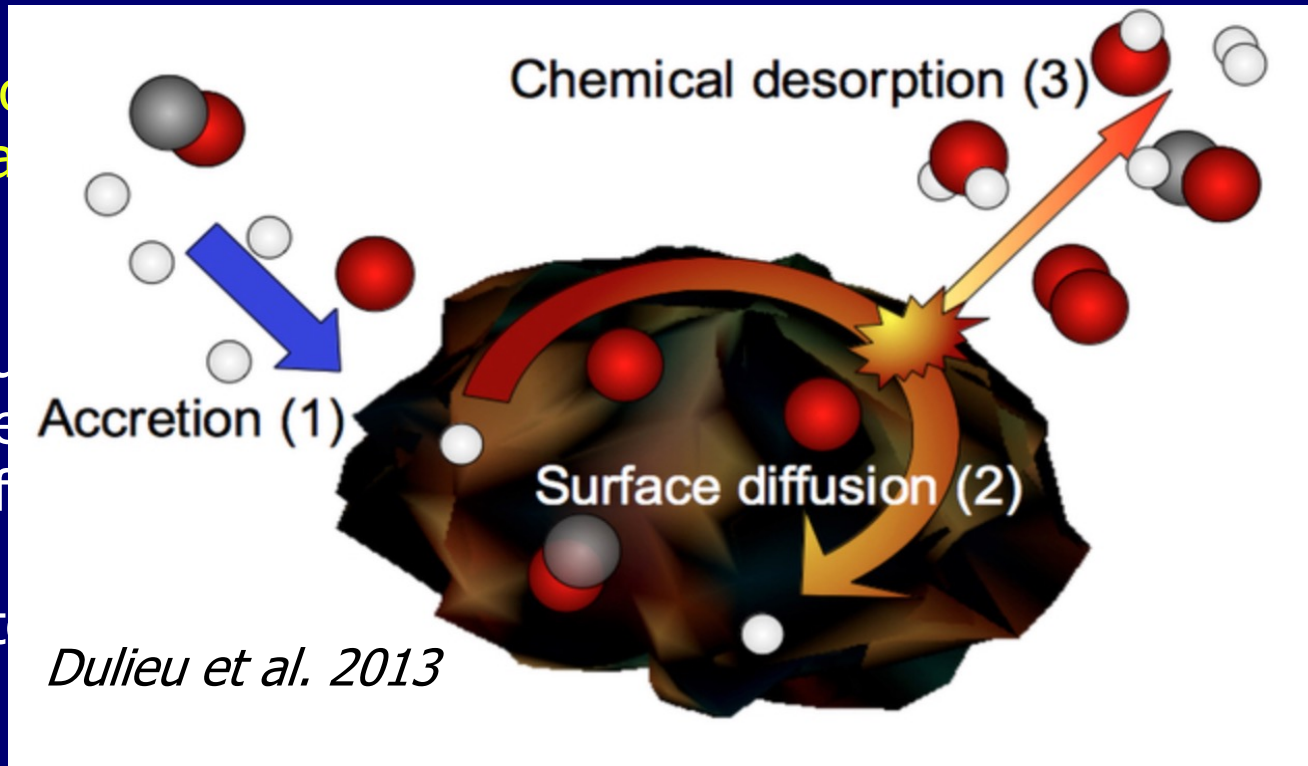


# Molecular ISM Basics

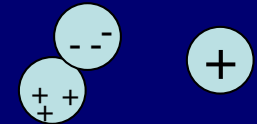
- Ion-molecule
- Neutral

- Ion induced  
--> creation  
--> efficient

- At low temperatures



important



of molecules.

- However, not enough ions to account for large  $H_2$  abundances  
--> grain surface chemistry important

- Simple molecules like CO or CS  $\rightarrow$  ion-molecule chemistry,
- More complex molecules  $\rightarrow$  grain surface chemistry important

# Molecules in Space

## Molecules in the Interstellar Medium or Circumstellar Shells (as of 09/2024)

2 atoms	3 atoms	4 atoms	5 atoms	6 atoms	7 atoms	8 atoms	9 atoms	10 atoms	11 atoms	12 atoms	>12 atoms
CH <sup>+</sup>	C <sub>3</sub> <sup>+</sup>	c-C <sub>3</sub> H	C <sub>5</sub> <sup>+</sup>	C <sub>2</sub> H	C <sub>2</sub> H	CH <sub>3</sub> C <sub>2</sub> N	CH <sub>3</sub> C <sub>2</sub> H	CH <sub>3</sub> C <sub>2</sub> N	HC <sub>2</sub> N	c-C <sub>6</sub> H <sub>6</sub> <sup>+</sup>	C <sub>60</sub> <sup>+</sup>
CH	C <sub>2</sub> H	I-C <sub>3</sub> H	C <sub>2</sub> H	I-H <sub>2</sub> C <sub>4</sub>	CH <sub>2</sub> CHCN	HC(O)OCH <sub>3</sub>	CH <sub>3</sub> CH <sub>2</sub> CN	(CH <sub>3</sub> ) <sub>2</sub> CO	CH <sub>3</sub> C <sub>6</sub> H	n-C <sub>3</sub> H <sub>7</sub> CN	C <sub>70</sub> <sup>+</sup>
CN	C <sub>2</sub> O	C <sub>3</sub> N	C <sub>4</sub> Si	C <sub>2</sub> H <sub>4</sub> <sup>+</sup>	CH <sub>3</sub> C <sub>2</sub> H	CH <sub>3</sub> COOH	(CH <sub>3</sub> ) <sub>2</sub> O	(CH <sub>2</sub> OH) <sub>2</sub>	C <sub>2</sub> H <sub>3</sub> OCHO	I-C <sub>3</sub> H <sub>7</sub> CN	C <sub>60</sub> <sup>++</sup>
OH	C <sub>2</sub> S	C <sub>3</sub> O	I-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> CN	HC <sub>2</sub> N	C <sub>7</sub> H	CH <sub>3</sub> CH <sub>2</sub> OH	CH <sub>3</sub> CH <sub>2</sub> CHO	CH <sub>3</sub> OC(O)CH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub> OCH <sub>3</sub>	c-C <sub>6</sub> H <sub>5</sub> CN
CO	CH <sub>2</sub>	C <sub>3</sub> S	c-C <sub>3</sub> H <sub>2</sub>	CH <sub>3</sub> NC	CH <sub>3</sub> CHO	C <sub>6</sub> H <sub>2</sub>	HC <sub>2</sub> N	CH <sub>3</sub> CHCH <sub>2</sub> O	CH <sub>3</sub> C(O)CH <sub>2</sub> OH	1-c-C <sub>3</sub> H <sub>5</sub> CN	HC <sub>11</sub> N
H <sub>2</sub>	HCN	C <sub>2</sub> H <sub>2</sub> <sup>+</sup>	H <sub>2</sub> CCN	CH <sub>3</sub> OH	CH <sub>3</sub> NH <sub>2</sub>	CH <sub>2</sub> OHCHO	C <sub>6</sub> H	CH <sub>3</sub> OCH <sub>2</sub> OH	c-C <sub>6</sub> H <sub>6</sub>	2-c-C <sub>3</sub> H <sub>5</sub> CN	1-C <sub>10</sub> H <sub>7</sub> CN
SiO	HCO	NH <sub>3</sub>	CH <sub>4</sub> <sup>+</sup>	CH <sub>3</sub> SH	c-C <sub>2</sub> H <sub>4</sub> O	I-HC <sub>6</sub> H <sup>+</sup>	CH <sub>3</sub> C(O)NH <sub>2</sub>	c-C <sub>6</sub> H <sub>4</sub>	HOCH <sub>2</sub> CH <sub>2</sub> NH <sub>2</sub>	CH <sub>3</sub> C <sub>7</sub> N (?)	2-C <sub>10</sub> H <sub>7</sub> CN
CS	HCO <sup>+</sup>	HCCN	HC <sub>3</sub> N	HC <sub>2</sub> NH <sup>+</sup>	H <sub>2</sub> CCHOH	CH <sub>2</sub> CHCHO	C <sub>6</sub> H <sup>+</sup>	H <sub>2</sub> CCCHC <sub>3</sub> N	H <sub>2</sub> CCCHC <sub>4</sub> H	n-C <sub>3</sub> H <sub>7</sub> OH	c-C <sub>6</sub> H <sub>5</sub>
SO	HCS <sup>+</sup>	HCNH <sup>+</sup>	HCCNC	HCCCHO	C <sub>6</sub> H <sup>+</sup>	CH <sub>2</sub> CCHCN	C <sub>3</sub> H <sub>6</sub>	C <sub>2</sub> H <sub>6</sub> NCO	C <sub>10</sub> H <sup>+</sup> (2023)	I-C <sub>3</sub> H <sub>5</sub> OH	1-c-C <sub>3</sub> H <sub>5</sub> CCH
SIS	HOC <sup>+</sup>	HNCO	HCOOH	NH <sub>2</sub> CHO	CH <sub>3</sub> NC	H <sub>2</sub> NCH <sub>2</sub> CN	CH <sub>3</sub> CH <sub>2</sub> SH	C <sub>2</sub> H <sub>6</sub> NH <sub>2</sub> (?)	H <sub>2</sub> C(CH <sub>3</sub> ) <sub>3</sub> CN (2023)	(CH <sub>3</sub> ) <sub>2</sub> C=CH <sub>2</sub> (2023)	2-c-C <sub>3</sub> H <sub>5</sub> CCH
NS	H <sub>2</sub> O	HNCS	H <sub>2</sub> CNH	C <sub>3</sub> N	HC <sub>2</sub> O	CH <sub>3</sub> CHNH	CH <sub>3</sub> NHCHO	HC <sub>7</sub> NH <sup>+</sup>			c-C <sub>3</sub> H <sub>4</sub> CCH <sub>2</sub>
C <sub>2</sub> <sup>++</sup>	H <sub>2</sub> S	HOCCO <sup>+</sup>	H <sub>2</sub> C <sub>2</sub> O	I-HC <sub>4</sub> H <sup>+</sup>	HOCH <sub>2</sub> CN	CH <sub>3</sub> SiH <sub>3</sub>	HC <sub>7</sub> O	E-CH <sub>3</sub> CHCHCN			2-C <sub>6</sub> H <sub>7</sub> CN
NO	HNC	H <sub>2</sub> CO	H <sub>2</sub> CNCN	I-HC <sub>4</sub> N	HCCCHNH	H <sub>2</sub> NC(O)NH <sub>2</sub>	HCCCHCHCN	Z-CH <sub>3</sub> CHCHCN			C <sub>6</sub> H <sub>3</sub> CCH (2023)
HCl	HNO	H <sub>2</sub> CN	HNC <sub>3</sub>	c-H <sub>2</sub> C <sub>3</sub> O	HC <sub>2</sub> CN	HCCCH <sub>2</sub> CN	H <sub>2</sub> CCHC <sub>3</sub> N	CH <sub>3</sub> C(CN)CH <sub>3</sub>			CH <sub>3</sub> OCH <sub>2</sub> CH <sub>2</sub> OH (2024)
NaCl	MgCN	H <sub>2</sub> CS	SiH <sub>4</sub> <sup>+</sup>	H <sub>2</sub> CCNH	c-C <sub>3</sub> HCCH	HC <sub>6</sub> NH <sup>+</sup>	H <sub>2</sub> CCCHCCH	CH <sub>2</sub> CHCH <sub>2</sub> CN			
KCl	MgNC	H <sub>3</sub> O <sup>+</sup>	H <sub>2</sub> COH <sup>+</sup>	C <sub>3</sub> N <sup>-</sup>	I-H <sub>2</sub> C <sub>5</sub>	CH <sub>2</sub> CHCCH	HOCHCHCHO (2024)	HOCH <sub>2</sub> C(O)NH <sub>2</sub> (2023)			
AlCl	N <sub>2</sub> H <sup>+</sup>	c-SiC <sub>3</sub>	C <sub>4</sub> H <sup>+</sup>	HNCHCN	MgC <sub>3</sub> N	MgC <sub>6</sub> H	HC <sub>7</sub> N <sup>+</sup> (2024)	CH <sub>3</sub> CH <sub>2</sub> CCH (2024)			
AlF	N <sub>2</sub> O	CH <sub>3</sub> <sup>+</sup>	HC(O)CN	SiH <sub>3</sub> CN	CH <sub>2</sub> C <sub>3</sub> N	C <sub>2</sub> H <sub>3</sub> NH <sub>2</sub>	CH <sub>2</sub> (CCH <sub>2</sub> ) <sub>2</sub> (2024)				
PN	NaCN	C <sub>3</sub> N <sup>-</sup>	HNCNH	C <sub>2</sub> S	NC <sub>4</sub> NH <sup>+</sup> (2023)	(CHOH) <sub>2</sub>					
SiC	OCS	PH <sub>3</sub>	CH <sub>3</sub> O	MgC <sub>2</sub> H	MgC <sub>3</sub> N <sup>+</sup> (2023)	HC <sub>2</sub> (H)C <sub>4</sub>					
CP	SO <sub>2</sub>	HCNO	NH <sub>4</sub> <sup>+</sup>	CH <sub>3</sub> CO <sup>+</sup>	HC <sub>6</sub> N <sup>+</sup> (2024)	C <sub>7</sub> N <sup>-</sup> (2023)					
NH	c-SiC <sub>2</sub>	HOCN	H <sub>2</sub> NCO <sup>+</sup>	C <sub>3</sub> H <sub>3</sub>	HNC <sub>5</sub> (2024)	CH <sub>3</sub> CHCO (2023)					
SIN	CO <sub>2</sub> <sup>+</sup>	HSCN	NCNCNH <sup>+</sup>	H <sub>2</sub> C <sub>3</sub> S	CH <sub>2</sub> (CN) <sub>2</sub> (2024)	MgC <sub>6</sub> H <sup>+</sup> (2023)					
SO <sup>+</sup>	NH <sub>2</sub>	H <sub>2</sub> O <sub>2</sub>	CH <sub>2</sub> Cl	HCCCHS		Z-(CH <sub>2</sub> ) <sub>2</sub> (CN) <sub>2</sub> (2024)					
CO <sup>+</sup>	H <sub>3</sub> <sup>++</sup>	C <sub>3</sub> H <sup>+</sup>	MgC <sub>3</sub> N	C <sub>5</sub> O							
HF	SiCN	HMgNC	NH <sub>2</sub> OH	C <sub>6</sub> H <sup>+</sup>							
SiH?	AlNC	HCCO	HC <sub>3</sub> O <sup>+</sup>	HCCNCH <sup>+</sup>							
FeO?	SINC	CNCN	HC <sub>3</sub> S <sup>+</sup>	c-C <sub>3</sub> C <sub>2</sub> H							
O <sub>2</sub>	HCP	HONO	H <sub>2</sub> C <sub>2</sub> S	HC <sub>4</sub> S							
CF <sup>+</sup>	CCP	MgC <sub>2</sub> H	C <sub>4</sub> S	HMgC <sub>3</sub> N (2023)							
PO	AlOH	HCCS	HC(O)SH	MgC <sub>4</sub> H <sup>+</sup> (2023)							
AlO	H <sub>2</sub> O <sup>+</sup>	HNCN	HC(S)CN	H <sub>2</sub> C <sub>3</sub> H <sup>+</sup> (2023)							
OH <sup>+</sup>	H <sub>2</sub> Cl <sup>+</sup>	H <sub>2</sub> NC	HCCCO	H <sub>2</sub> C <sub>3</sub> N (2023)							
CN <sup>-</sup>	KCN	HCOS <sup>+</sup>	NaCCCN (2023)	(HO) <sub>2</sub> CO (2023)							
SH <sup>+</sup>	FeCN	CH <sub>3</sub> <sup>+</sup> (2023)	MgC <sub>3</sub> N <sup>+</sup> (2023)	H <sub>2</sub> CNCN (2024)							
SH	HO <sub>2</sub>	HCNS (2024)	HC <sub>3</sub> N <sup>+</sup> (2024)	NCHCCS (2024)							
HCl <sup>+</sup>	TiO <sub>2</sub>	HOCS <sup>+</sup> (2024)	HC <sub>3</sub> S (2024)								
TiO	C <sub>2</sub> N	HNSO (2024)	NC <sub>3</sub> S (2024)								
ArH <sup>+</sup>	Si <sub>2</sub> C										
N <sub>2</sub>	HS <sub>2</sub>										
NO <sup>+</sup>	HCS										
NS <sup>+</sup>	HSC										
HeH <sup>+</sup>	NCO										
PO <sup>+</sup>	CaNC										
SIP ?	NCS										
FeC (2023)	MgC <sub>2</sub>										
	HSO (2023)										
	CaC <sub>2</sub> (2024)										

www.cdms.de

More than 320 detected interstellar molecules as of October 2024 ([www.cdms.de](http://www.cdms.de)).  
74 molecular detection in extragalactic systems.

# A few important molecules

Mol.	Trans.	Abund.	Crit. dens. [cm <sup>-3</sup> ]	Comments
H <sub>2</sub>	1-0 S(1)	1	8x10 <sup>7</sup>	Shock tracer
CO	J=1-0	8x10 <sup>-5</sup>	3x10 <sup>3</sup>	Low-density probe
OH	<sup>2</sup> Π <sub>3/2</sub> ; J=3/2	3x10 <sup>-7</sup>	1x10 <sup>0</sup>	Magnetic field probe (Zeeman)
NH <sub>3</sub>	J,K=1,1	2x10 <sup>-8</sup>	2x10 <sup>4</sup>	Temperature probe
CS	J=2-1	1x10 <sup>-8</sup>	4x10 <sup>5</sup>	High-density probe
SiO	J=2-1		6x10 <sup>5</sup>	Outflow shock tracer
H <sub>2</sub> O	6 <sub>16</sub> -5 <sub>23</sub>		1x10 <sup>3</sup>	Maser
H <sub>2</sub> O	1 <sub>10</sub> -1 <sub>11</sub>	<7x10 <sup>-8</sup>	2x10 <sup>7</sup>	Warm gas probe
CH <sub>3</sub> OH	7-6	1x10 <sup>-7</sup>	1x10 <sup>5</sup>	Dense gas/temperature probe
CH <sub>3</sub> CN	19-18	2x10 <sup>-8</sup>	2x10 <sup>7</sup>	Temperature probe in Hot Cores

Crit. Dens.:  $n_{\text{crit}} \sim A/\gamma$



# Basics IV

## Depletion of molecules on dust grains

In molecule's ref. frame, grains are moving at  $v_{\text{therm}}$  relative to molecules

$$E = 1/2 m v_{\text{therm}}^2 = 3/2 k_b T \Rightarrow v_{\text{therm}} = (3k_b T/m)^{1/2}$$

n grains sweeps out cylindrical volume in time  $\Delta t$  of  
 $n(\pi a^2)v_{\text{therm}}\Delta t$  (a: grain radius)

Probability of molecule in volume V to be struck by grain in time  $\Delta t$

$$P(\Delta t) = n(\pi a^2)v_{\text{therm}}\Delta t / V$$

Hence the collision time  $t_{\text{coll}}$  (for  $P(\Delta t)V = 1$ )

$$t_{\text{coll}} = 1/(n(\pi a^2)v_{\text{therm}}) = 1/(n_H \Sigma v_{\text{therm}}) \quad (n_H: \text{density}; \Sigma: \text{grain cross section})$$

For example CS:  $v_{\text{therm}} \sim 5 \times 10^3 \text{ cm s}^{-1}$  at 10K,  $n_H \sim 10^4 \text{ cm}^{-3}$ ,  $\Sigma \sim 10^{-21} \text{ cm}^2$

$$t_{\text{coll}} \sim 6 \times 10^5 \text{ yr}$$

Depletion time-scale very short

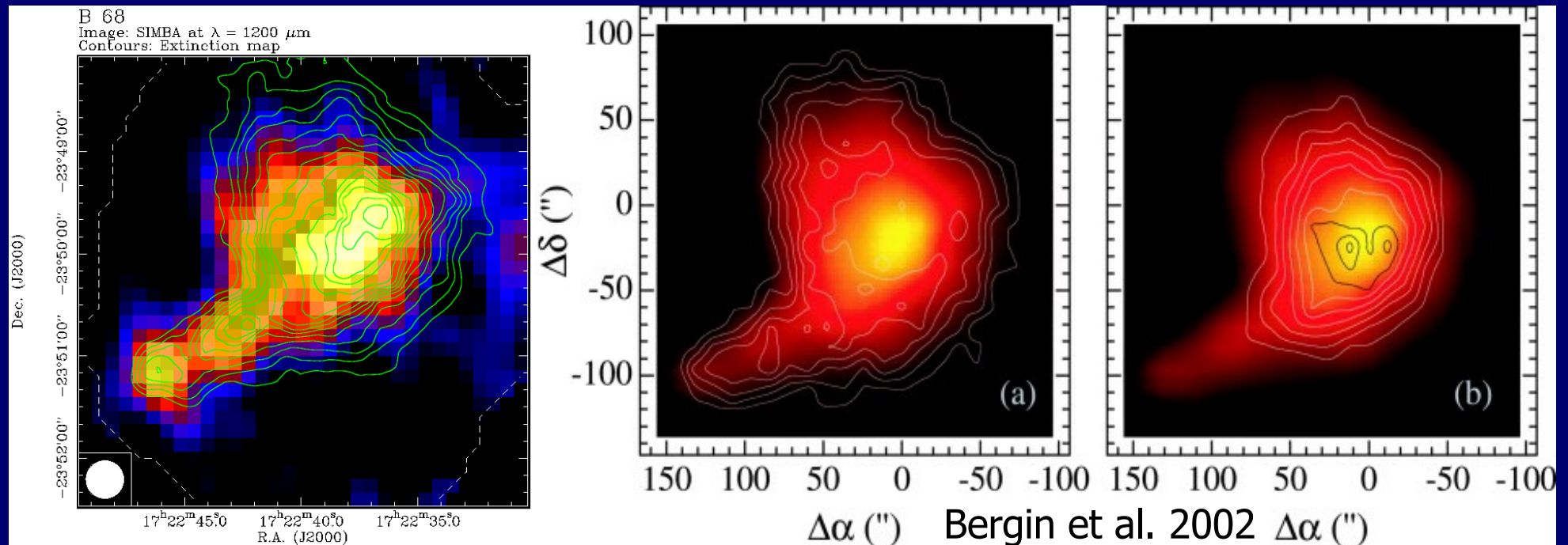
--> mechanisms for re-injecting molecules from grains important

# Depletion example

1.2 mm Dust Continuum

$C^{18}O$

$N_2H^+$



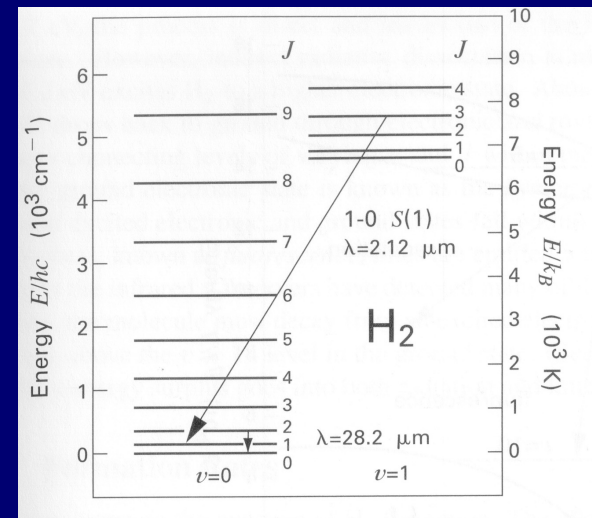
Possible mechanisms working against depletion:

- UV radiation (not working in dense cores)
- In small grain, heat from chemical grain surface reactions could raise temperature
- Kelvin-Helmholtz contraction and energy
- Ignited central protostar
- Shocks
- ...

# Molecular Hydrogen (H<sub>2</sub>)

- H<sub>2</sub> consists of 2 identical atoms → no dipole moment
- Rotationally excited H<sub>2</sub> has allowed quadrupole transitions  $\Delta J = 2$   
→ lowest rotational transition J=2-0 has energy change of 510 K

- Rotational energy for H<sub>2</sub>:  
Classical mechanics:  $E_{\text{rot}} = J^2/2I$   
(J: Angular momentum; I: Moment of inertia)  
→ Small moment of inertia ( $I=mr^2$ )  
→ large spread of energy levels



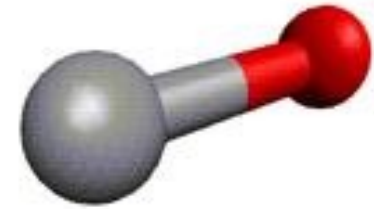
*Stahler & Palla 2004*

- Cold clouds have to be observed other ways, e.g., CO



# Carbon monoxide (CO)

- Forms through gas phase reactions.
- Strong binding energy of 11.1 eV  
→ prevents much further destruction (self-shielding).



<http://www.cdms.de>

- Permanent dipole moment → strong emission at (sub)mm wavelengths.

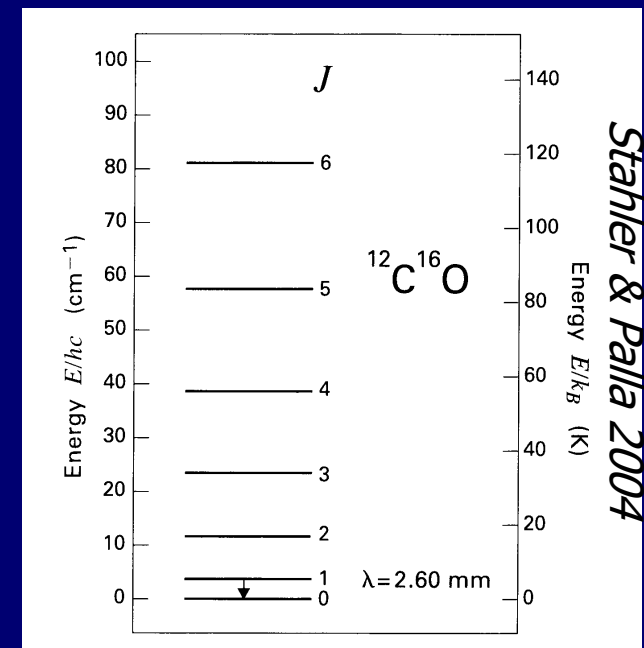
- Larger moment of inertia than H<sub>2</sub>.  
→ more closely spaced rotational ladder,  
J=1 level at 4.8x10<sup>-4</sup>eV or 5.5K above ground

- In molecular clouds excitation mainly via collisions with H<sub>2</sub>.
- Critical density for thermodynamic equilibrium with H<sub>2</sub>  $n_{\text{crit}} = A/\gamma \sim 3 \times 10^3 \text{cm}^{-3}$ .  
(A: Einstein A coefficient;  $\gamma$ : collision rate with H<sub>2</sub>)

- The level population follows a Boltzmann-law:

$$n_{J+1}/n_J = g_{J+1}/g_J \exp(-\Delta E/k_B T_{\text{ex}}) \quad (\text{for CO, the statistical weights } g_J = 2J + 1)$$

Excitation temperature  $T_{\text{ex}}$  measure for the level populations and equals the kinetic temperature  $T_{\text{kin}}$  if densities are  $> n_{\text{crit}}$ .



# Topics today

- The ISM, molecules and depletion
- Heating and cooling
- Radiation transfer and column density determination

# Heating processes

UV radiation from stars

Energy injection from supernovae

Energy injection from outflows/jets

---

- Cosmic rays interact with HI and H<sub>2</sub>  
(mainly relativistic protons accelerated within magnetized shocks produced by supernova-remnant--molecular cloud interactions)



- Interstellar radiation (diffuse field permeating interstellar space)  
Mainly dissociates carbon (lower ionization potential than H<sub>2</sub>)



- Photoelectric heating:
  - Heats grains that re-radiate in infrared regime
  - UV photons eject e<sup>-</sup> from dust
  - e<sup>-</sup> heat surrounding gas via collisions



# Cooling processes

- H & H<sub>2</sub> no dipole moment → no efficient coolant in cold molecular cloud  
→ other coolants needed

--> Hydrogen collides with ambient atoms/molecules/grains  
→ Cooling via these secondary constituents.



At higher densities other molecules come into play, e.g., H<sub>2</sub>O.

→ CO the most effective coolant in molecular clouds.

- Collisions with gas atoms/molecules cause lattice vibrations on grain surfaces, that decay through the emission of infrared photons.

→ dust very efficient coolant



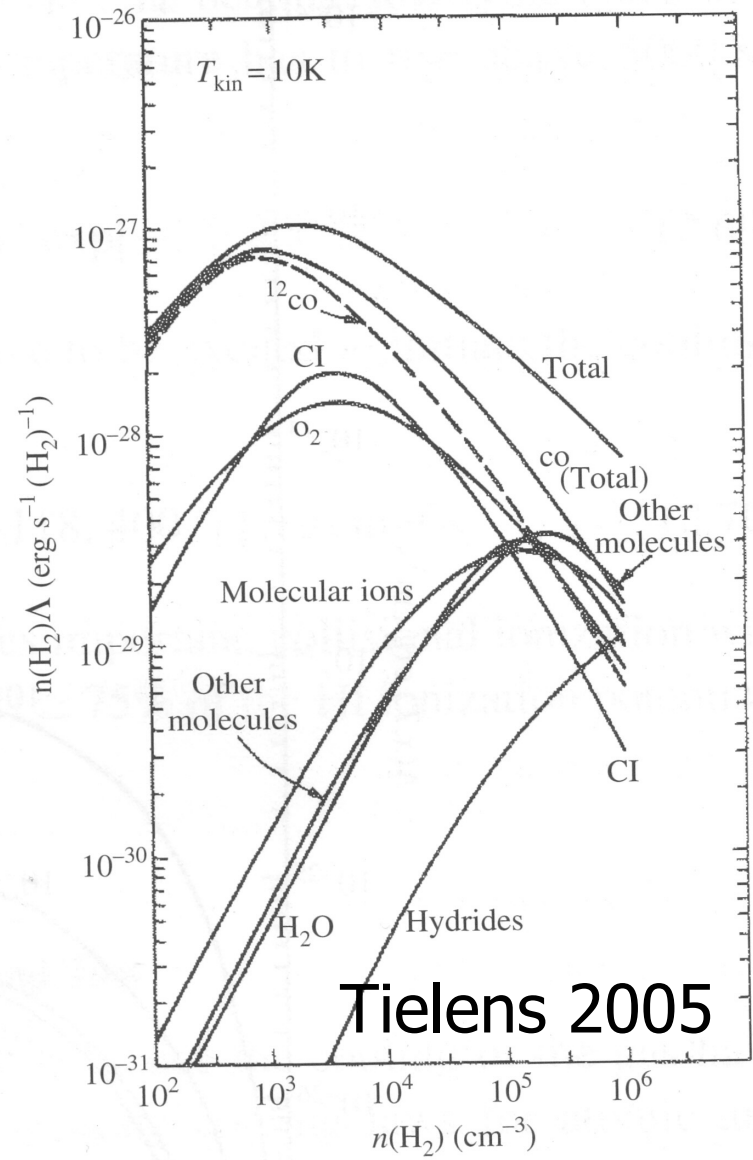
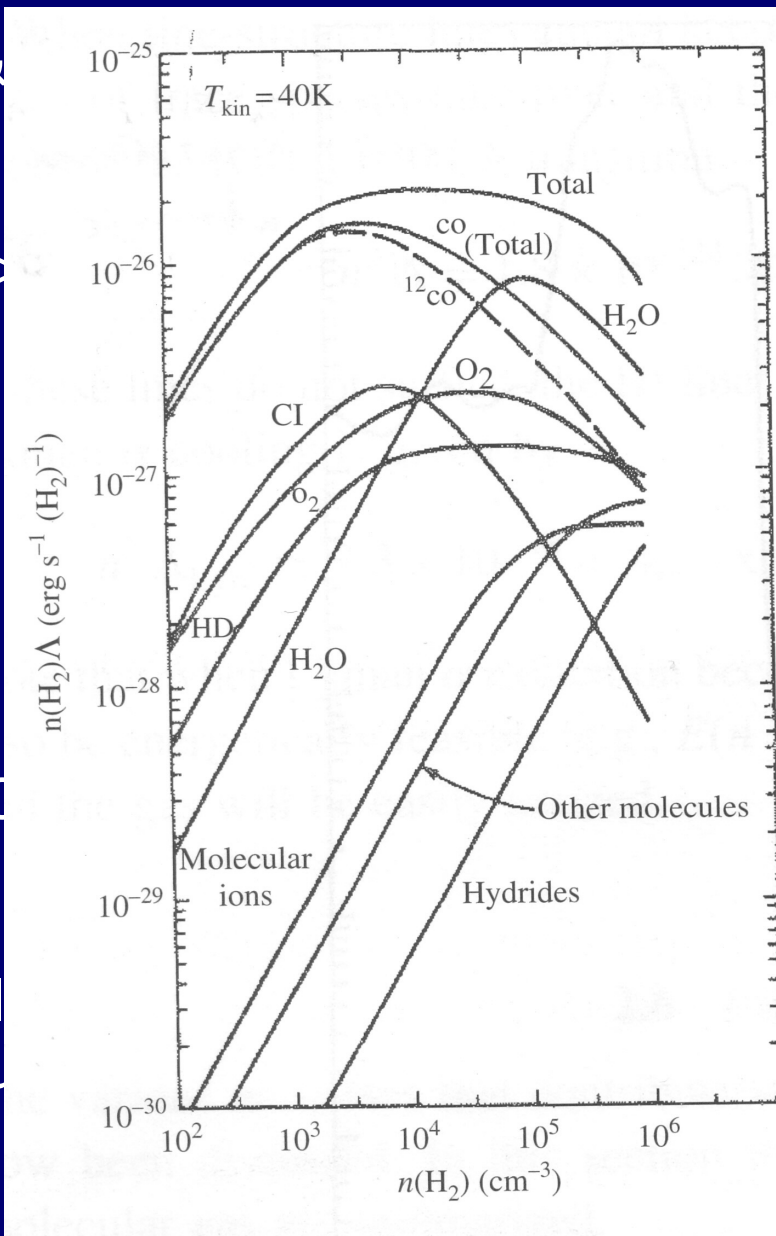
# Cooling processes

- H &

-

→ CO

- Coll  
su



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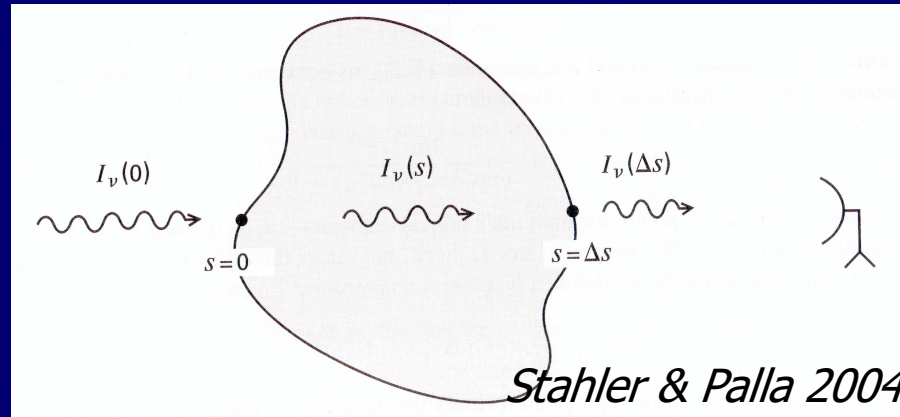
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# Topics today

- The ISM, molecules and depletion
- Heating and cooling
- Radiation transfer and column density determination



# Radiation transfer I



$$dI_\nu = -\kappa_\nu I_{\nu,0} ds + \varepsilon_\nu ds$$

with the opacity

$$d\tau_\nu = -\kappa_\nu ds$$

and the source function

$$S_\nu = \varepsilon_\nu / \kappa_\nu$$

$$\Rightarrow dI_\nu / d\tau_\nu = I_{\nu,0} - S_\nu$$

Assuming a spatially constant source function  $\rightarrow$  radiation transfer equation

$$\Rightarrow I_\nu = S_\nu (1 - e^{-\tau(\nu)}) + I_{\nu,0} e^{-\tau(\nu)}$$

$\kappa$ : absorption coef.  
 $\varepsilon$ : emission coef.

# Radiation transfer II

The excitation temperature  $T_{\text{ex}}$  is defined via a Boltzmann distribution as

$$n_J/n_{J-1} = g_J/g_{J-1} \exp(-h\nu/kT_{\text{ex}})$$

with  $n_J$  and  $g_J$  the number density and statistical weights.

In case of rotational transitions

$$g_J = 2J + 1$$

J: rot. quantum  
number

In thermal equilibrium

$$T_{\text{ex}} = T_{\text{kin}}$$

In a uniform molecular cloud the source function  $S_\nu$  equals Planck function

$$S_\nu = B_\nu(T_{\text{ex}}) = 2h\nu^3/c^2 (\exp(h\nu/kT_{\text{ex}}) - 1)^{-1}$$



# Radiation transfer III

Then the radiation transfer equation

$$\Rightarrow I_\nu = B_\nu (T_{\text{ex}}) (1 - e^{-\tau(\nu)}) + I_{\nu,0} e^{-\tau(\nu)}$$

In the Rayleigh-Jeans limits ( $h\nu \ll kT$ ) B equals

$$B = (2k\nu^2/c^2) T \quad (\text{def. } \rightarrow T = c^2/(2k\nu^2) I_\nu)$$

And the radiation transfer equation using now the radiation temperature is

$$T_r = J_\nu (T_{\text{ex}}) (1 - e^{-\tau(\nu)}) + J_{\nu,0} (T_{\text{bg}}) e^{-\tau(\nu)}$$

with

$$J_\nu = h\nu/k (\exp(h\nu/kT) - 1)^{-1}$$

# Molecular column densities I

To derive molecular column densities, 3 quantities are important:

- 1) Intensity  $T$  of the line
- 2) Optical depth  $\tau$  of the line (observe isotopologues or hyperfine structure)
- 3) Partition function  $Q$

The optical depth  $\tau$  of a molecular transition can be expressed like

$$\tau = \frac{c^2}{8\pi\nu^2} A_{ul} N_u (\exp(h\nu/kT) - 1) \phi$$

with the Einstein  $A_{ul}$  coefficient

$$A_{ul} = \frac{64\pi^4\nu^3}{3c^3h} \mu^2 J_u / (2J_u - 1)$$

and the line form function  $\phi$

$$\phi = \frac{c}{\nu} \frac{2\sqrt{\ln 2}}{\sqrt{\pi} \Delta\nu}$$

# Molecular column densities II

Using furthermore the radiation transfer eq. ignoring the background

$$T = J_\nu(T_{\text{ex}}) \tau (1 - e^{-\tau}) / \tau$$

And solving  $\tau$ -equation for  $N_u$ , one gets

$$N_u = 3k/8\pi^3 \nu^{-1} / \mu^2 (2J_u - 1) / J_u \tau / (1 - e^{-\tau}) (T \Delta \nu \sqrt{\pi}) / (2\sqrt{\ln 2})$$

The last expression equals the integral  $\int T dv$ .

$$\rightarrow N_u \sim \tau / (1 - e^{-\tau}) \int T dv$$

The column density in the upper level  $N_u$  relates to the total column density  $N_{\text{tot}}$

$$N_{\text{tot}} = N_u / g_u \exp(E_u / kT) Q$$

For linear molecule like CO, partition function  $Q$  approximated:  $Q = kT/hB$ .  
( $B$ : rotational constant)

However, for more complex molecules  $Q$  can become very complicated.

# Conversion from CO to H<sub>2</sub> column densities

Classical way to derive conversion factors from CO to H<sub>2</sub> column densities:

- 1) Derive ratio between colour excess  $E_{B-V}$  and optical extinction  $A_V$   
$$A_V = 3.1 E_{B-V} \quad (\text{Savage and Mathis, 1979})$$
  - 2) The ratio  $N(\text{H}_2)/E_{B-V}$ : One can measure the H<sub>2</sub> column density, e.g., directly from UV Absorption lines.
  - 3) The ratio  $N(\text{CO})/A_V$ : In regions of molecular gas emission, one can estimate  $A_V$  by star counts in the Infrared regime
- ⇒ Combining these three ratios: CO → H<sub>2</sub> column densities.



# Summary

- Main tools: Spectral line emission and thermal emission and extinction from dust (more on dust next week)
- Molecules interesting for themselves and chemistry
- However, also extremely useful to trace physical processes.
- Molecules deplete on grains at low temperatures
- Discussed main cooling and heating processes
- Discussed basic line radiation transfer and column density determination

# Sternentstehung - Star Formation

Winter term 2024/2025

Henrik Beuther, Thomas Henning & Caroline Gieser

15.10	<i>Today: Introduction &amp; Overview</i>	(Beuther)
22.10	<i>Physical processes I</i>	(Beuther)
29.10	--	
<b>05.11</b>	<b>Physical processes II</b>	<b>(Beuther)</b>
12.11	Molecular clouds as birth places of stars	(Beuther)
19.11	Molecular clouds (cont.), Jeans Analysis	(Henning)
26.11	Collapse models I	(Beuther)
03.12	Collapse models II	(Beuther)
10.12	Protostellar evolution	(Gieser)
17.12	Pre-main sequence evolution & outflows/jets	(Henning)
07.01	Accretion disks I	(Henning)
14.01	Accretion disks II	(Henning)
21.01	High-mass star formation, clusters and the IMF	(Gieser)
28.01	Extragalactic star formation	(Henning)
04.02	Planetarium@HdA, outlook, questions	
11.02	Examination week, no star formation lecture	(Beuther, Gieser, Henning)

**Book: Stahler & Palla: The Formation of Stars, Wileys**

More Information and the current lecture files: [http://www.mpia.de/homes/beuther/lecture\\_ws2425.html](http://www.mpia.de/homes/beuther/lecture_ws2425.html)  
[beuther@mpia.de](mailto:beuther@mpia.de), [henning@mpia.de](mailto:henning@mpia.de), [gieser@mpia.de](mailto:gieser@mpia.de)

Heidelberg Joint Astronomical Colloquium Winter Semester 2024/25

Tuesday October 22nd Main Lecture Theatre, Philosophenweg 12, 16:30 CEST

**Leah Morabito**

**(University of Durham):**

**The highest resolution  
at the lowest  
frequencies: what  
LOFAR can tell  
us about active  
galactic nuclei**

leftover emission  
from jet activity

galaxy centre  
with black hole

jet launched from  
black hole

<https://www.physik.uni-heidelberg.de/hephysto/>

Host: Eduardo Banados ([banados@mpia.de](mailto:banados@mpia.de))

Image Credit: L.K. Morabito; LOFAR Legacy Survey



# Heidelberg Joint Astronomical Colloquium Winter Semester 2024/25

Tuesdays, 16:30 Main Lecture Theatre, Philosophenweg 12

- 22 Oct Leah Morabito** (University of Durham)  
*The highest resolution at the lowest frequencies: what LOFAR can tell us about AGNs*
- 29 Oct Joe Callingham** (SRON, Leiden)  
*Radio stars and exoplanets: Discovering the space weather of other worlds*
- 5 Nov Jenny Greene** (Princeton University)  
*To be announced*
- 12 Nov Elena Pancino** (INAF, Florence)  
*To be announced*
- 19 Nov Daisuke Kawata** (University College London)  
*The JASMINE mission*
- 26 Nov Tadafumi Matsuno** (Astronomisches Rechen-Institut, Heidelberg)  
*Chemical Enrichments in the Milky Way and it's Accreted Dwarf Galaxies*
- 3 Dec Daniele Huppenkothen** (SRON, Utrecht)  
*To be announced*
- 10 Dec Barbara Ercolano** (Universitäts-Sternwarte München)  
*The atmospheres of discs and planets*
- 17 Dec Aurora Simionescu** (Leiden Observatory)  
*The beating hearts of galaxies:  
supermassive black hole feedback probed by X-ray spectroscopy*
- 7 Jan Ilse De Looze** (University of Ghent)  
*To be announced*
- 14 Jan Caroline Heneka** (Institut für Theoretische Physik, Heidelberg)  
*The Universe in multi-color: Astronomy at the dawn of intensity mapping and AI*
- 21 Jan Sylvia Ekstroem** (University of Geneva)  
*To be announced*
- 28 Jan Martin Pessah** (Niels Bohr Institute, Copenhagen)  
*To be announced*
- 4 Feb Amelie Saintonge** (UC London / MPI für Radioastronomie, Bonn)  
*To be announced*

<https://www.physik.uni-heidelberg.de/hephysto/index.php?s=event@id=1>

Caption: JWST-NIRCam image of the SNR Cassiopeia A with filters color coded as F162M: Blue F356W: Green F444W: Red  
NASA, ESA, CSA, STScI, Danny Milisavljevic (Purdue University), Ilse De Looze (UGent), Tea Temim (Princeton University)

